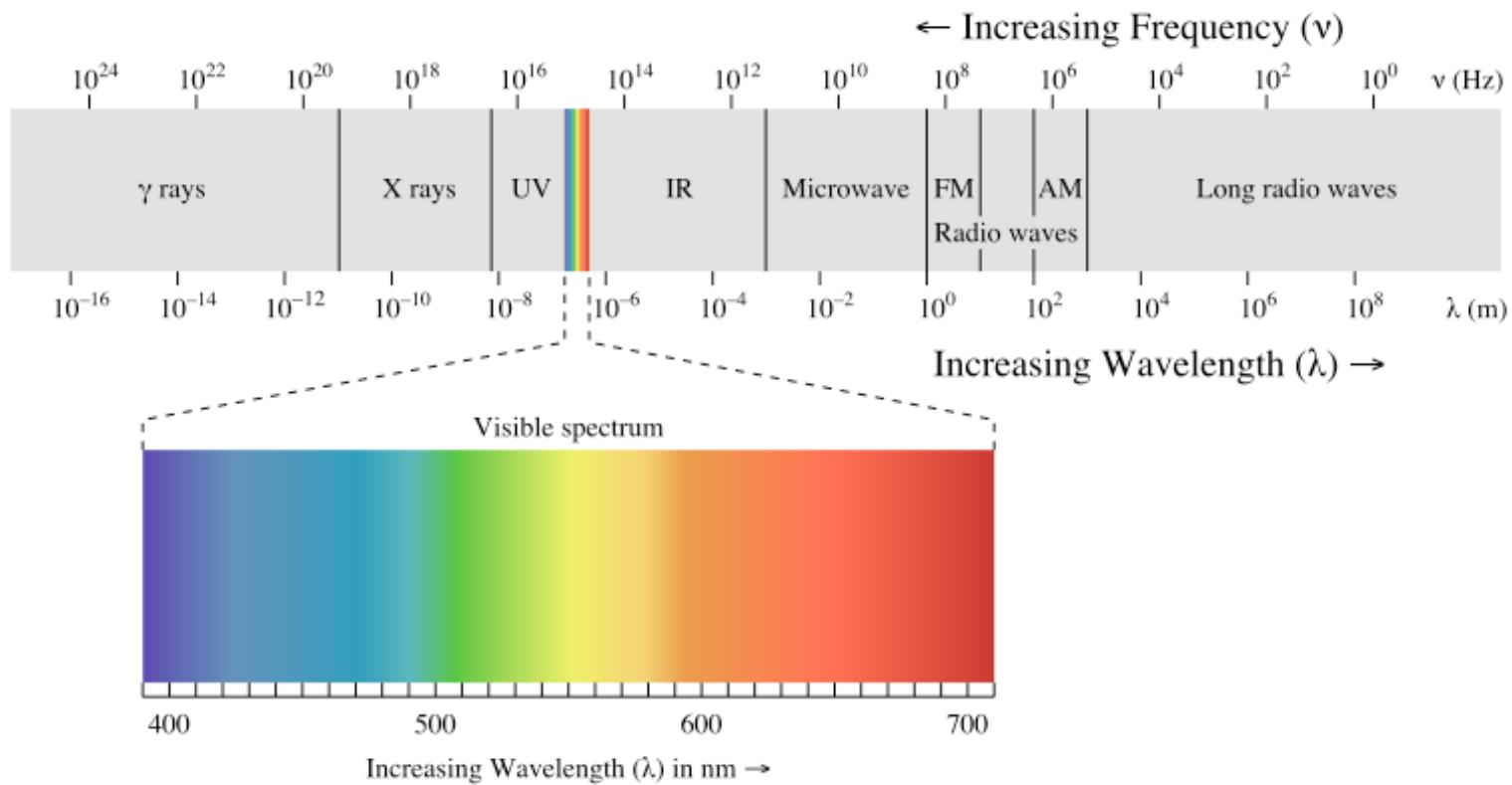


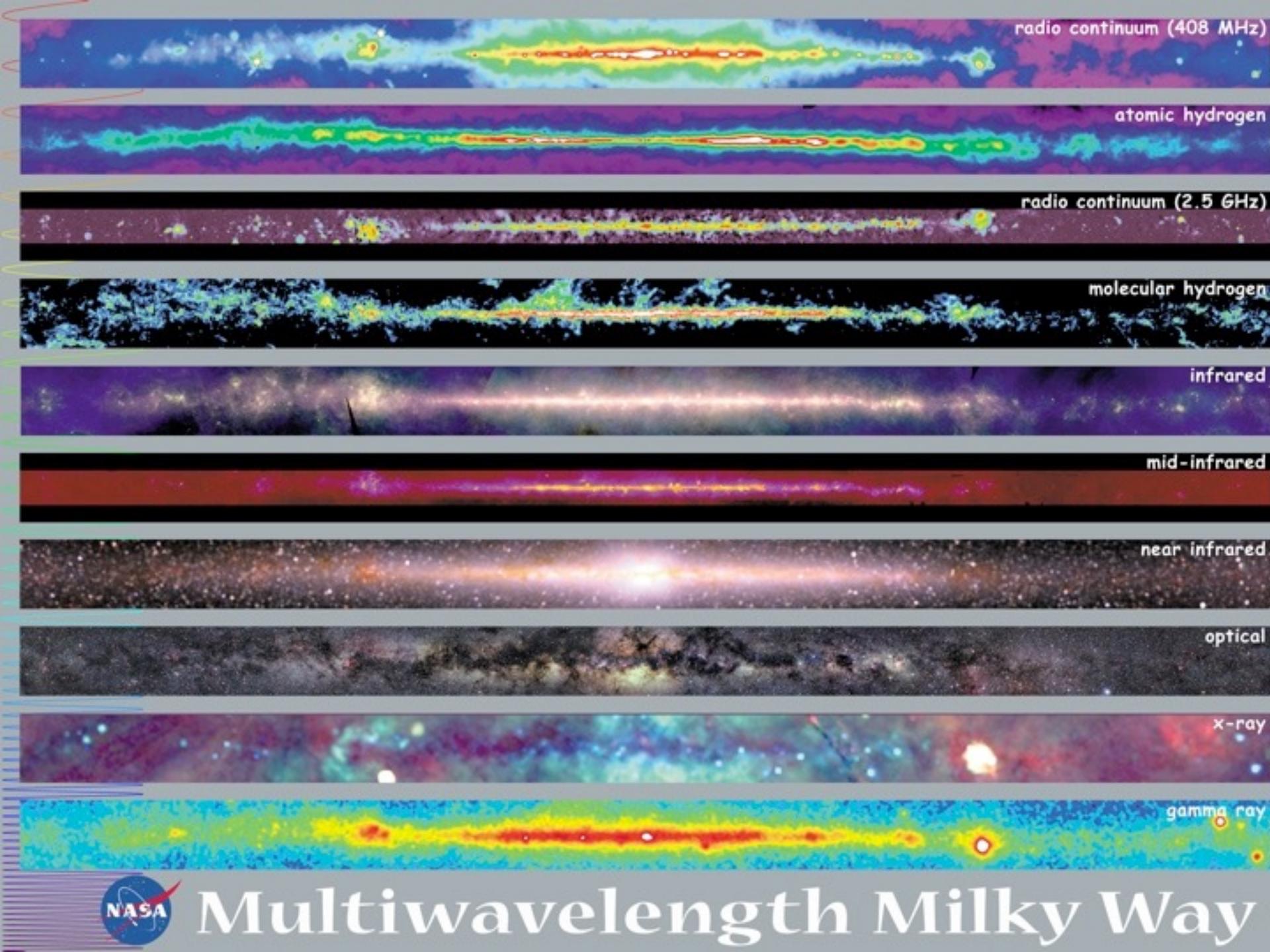
Introduction to *Radiation Transfer*

Hiroyuki Hirashita
(平下 博之)

1. Electromagnetic Radiation



All wavelengths are important in astronomy!



radio continuum (408 MHz)

atomic hydrogen

radio continuum (2.5 GHz)

molecular hydrogen

infrared

mid-infrared

near infrared

optical

x-ray

gamma ray

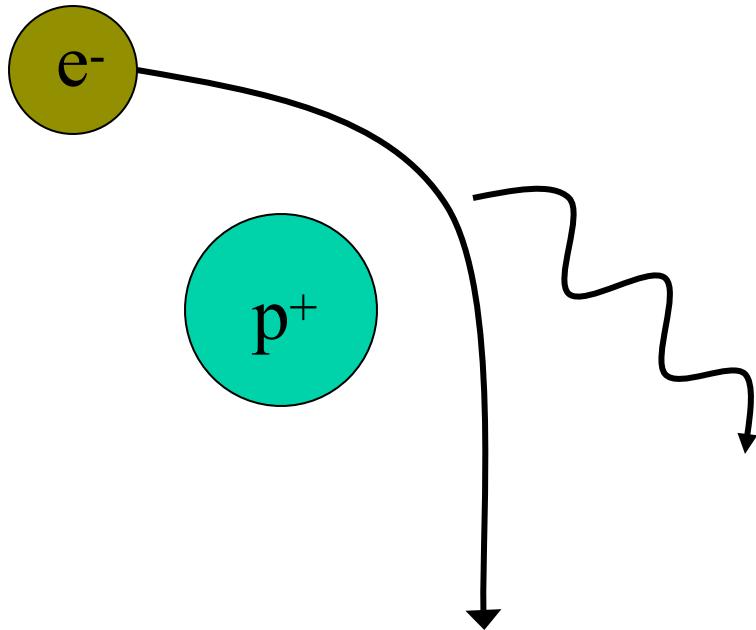


Multiwavelength Milky Way

Radiation Mechanisms

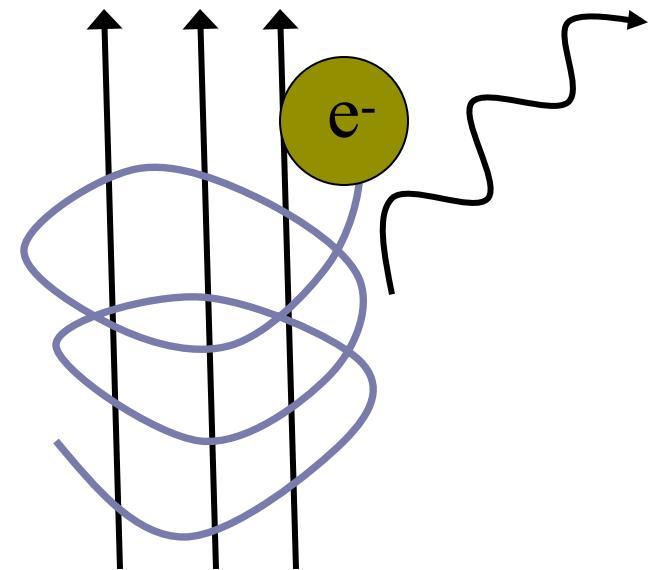
Radiation \Leftrightarrow Acceleration of electric charge

Bremsstrahlung (free-free)



Ionized gas:
e.g., H II region
Intracluster medium

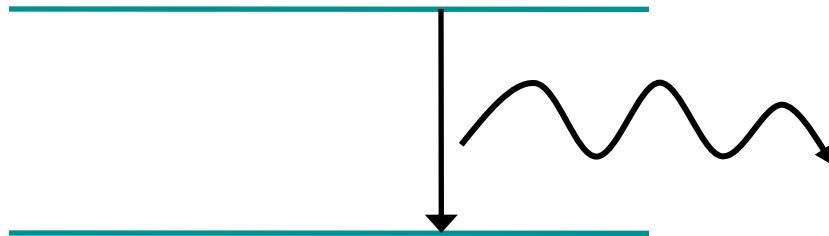
Synchrotron (magnetic
bremsstrahlung)



B Supernovae
AGN jet

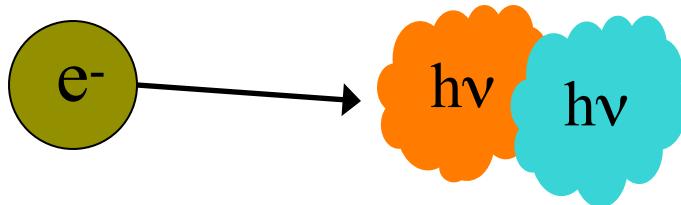
Radiation as Photons

Radiative transition



electric ~ UV, optical
molecular vibrational
~ near-IR
molecular rotational
~ radio

(Inverse) Compton Radiation

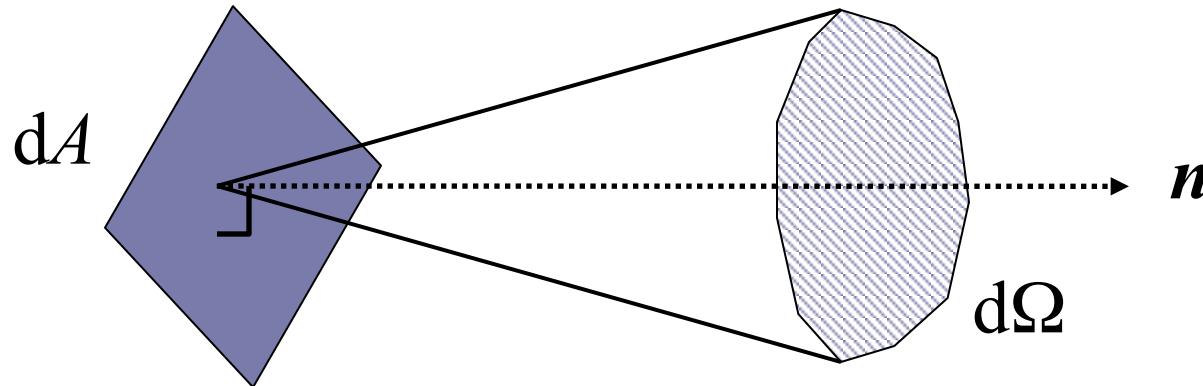


Intracluster medium
(Sunyaev-Zeldovich effect)
AGN jet, accretion disk

2. Radiation Transfer Equation

Definition of Intensity (energy flow on a line)

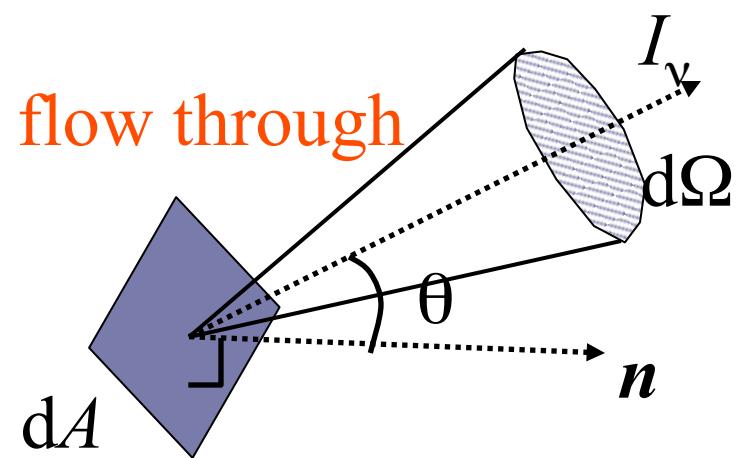
$$dE = I_\nu dA dt d\Omega dv \quad I_\nu [\text{erg/cm}^2/\text{s}/\text{sr}/\text{Hz}]$$



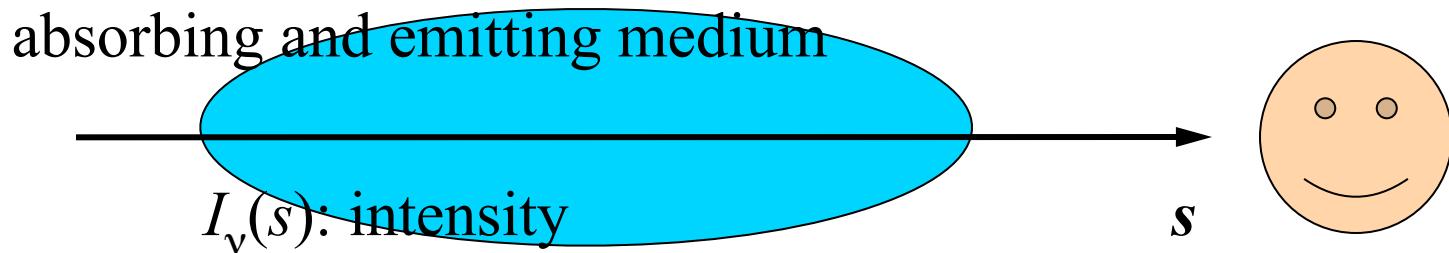
Definition of Flux (net energy flow through a surface)

$$F_\nu [\text{erg/cm}^2/\text{s}/\text{Hz}]$$

$$F_\nu = \oint I_\nu \cos \theta d\Omega$$



What is Radiation Transfer Equation?



$$\frac{dI_\nu}{ds} = -\kappa_\nu \rho I_\nu + j_\nu$$

Absorption

κ_ν [cm²/g]: mass absorption coefficient

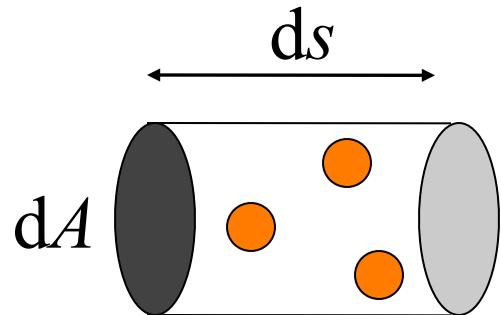
ρ [g/cm³]: density

Emission

j_ν [erg/cm³/s/Hz]

Homework

Absorption



Total absorbing area: $\sigma_v(n dA ds)$

n : number density of the absorber

σ_v : absorption cross section of the absorber particle

$$\Rightarrow -dI_v dA = I_v(n\sigma_v dA ds) \Rightarrow dI_v = -n\sigma_v I_v ds$$

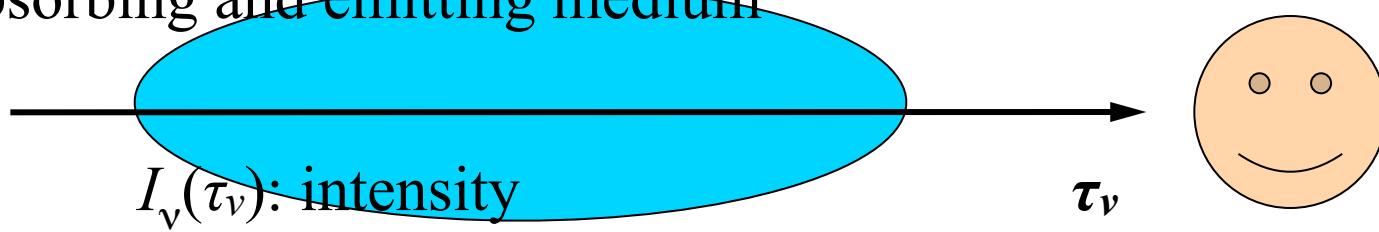
$$n\sigma_v = \rho\kappa_v$$

ρ : mass density ($= mn$; m is the mass of the absorber)

κ_v : mass absorption coefficient

Optical Depth

absorbing and emitting medium



$$\frac{dI_\nu}{ds} = -\kappa_\nu \rho I_\nu + j_\nu$$

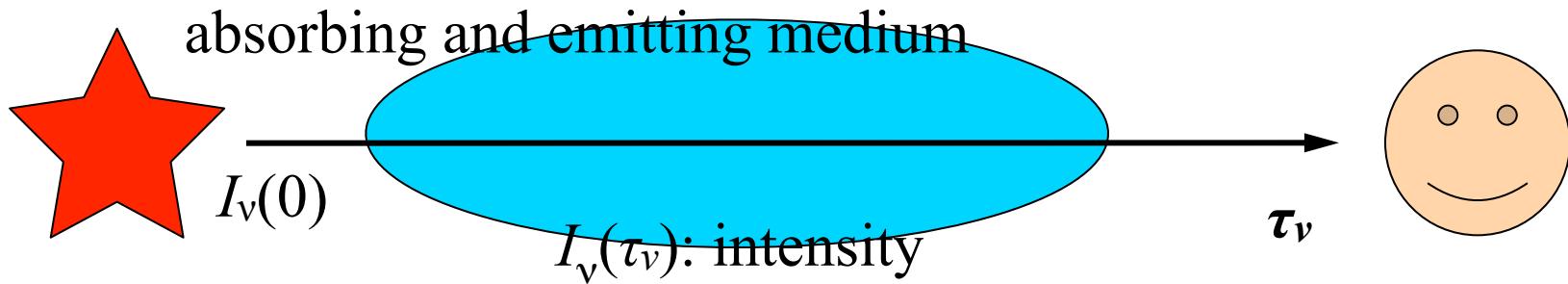
Optical depth: $d\tau_\nu = \rho \kappa_\nu ds$ ($= n \sigma_\nu ds$)

\propto “number of absorptions”

$$\frac{dI_\nu}{d\tau_\nu} = -I_\nu + S_\nu$$

$S_\nu = j_\nu / (\rho \kappa_\nu)$: source function

3. Simple Solutions



$$\frac{dI_\nu}{d\tau_\nu} = -I_\nu + S_\nu$$

For a constant source function:

$$I_\nu(\tau_\nu) = I_\nu(0)e^{-\tau_\nu} + S_\nu(1 - e^{-\tau_\nu})$$

Extinction of bkg light Emission

Extinction as Pure Absorption

With strong background source:

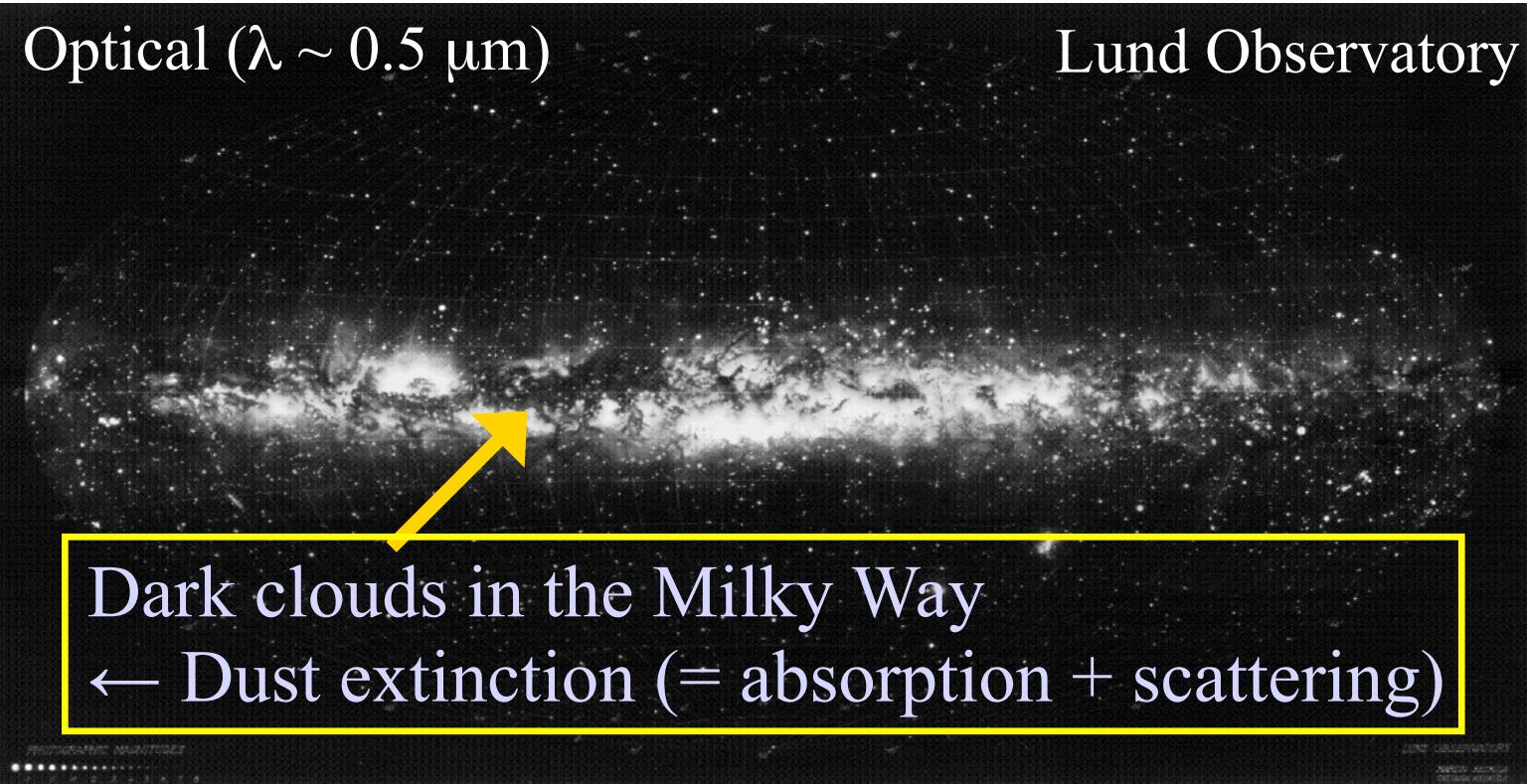
$$I_\nu(\tau_\nu) = I_\nu(0)e^{-\tau_\nu} + S_\nu(1 - e^{-\tau_\nu})$$

↓

$$I_\nu(\tau_\nu) = I_\nu(0)e^{-\tau_\nu}$$

Optical ($\lambda \sim 0.5 \text{ } \mu\text{m}$)

Lund Observatory



Optically Thick Case

Optically thick: $\tau_\nu \gg 1$

$$I_\nu(\tau_\nu) = I_\nu(0)e^{-\tau_\nu} + S_\nu(1 - e^{-\tau_\nu})$$



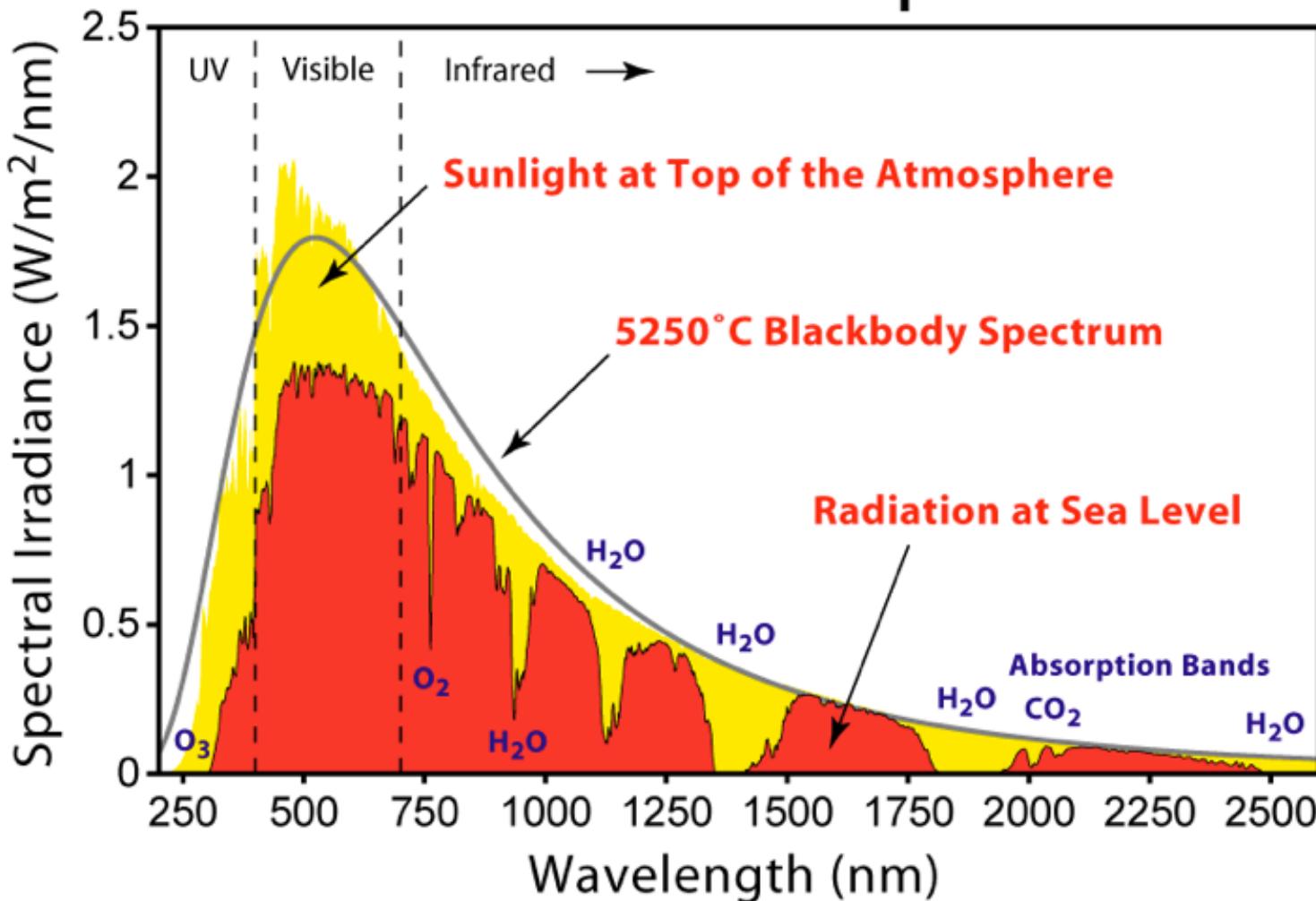
$$I_\nu = S_\nu$$

If the material is in **thermal equilibrium**, the resulting intensity should be the Planck function:

Local Thermal Equilibrium (LTE): $S_\nu(\tau_\nu) = B_\nu(T(\tau_\nu))$
($j_\nu = \rho \kappa B_\nu$): Kirchhoff's law.
$$B_\nu(T) = \frac{2h\nu^3}{c^2} \frac{1}{\exp(h\nu/kT) - 1}$$

Example of Optically Thick Radiation

Solar Radiation Spectrum



Optically Thin Emission

Optically thin: $\tau_\nu \ll 1$

Without background light $I_\nu(\tau_\nu) = S_\nu(1 - e^{-\tau_\nu})$



$$I_\nu(\tau_\nu) = \tau_\nu S_\nu$$

column density

For LTE $I_\nu(\tau_\nu) = \tau_\nu B_\nu = \kappa_\nu \rho L B_\nu(T)$

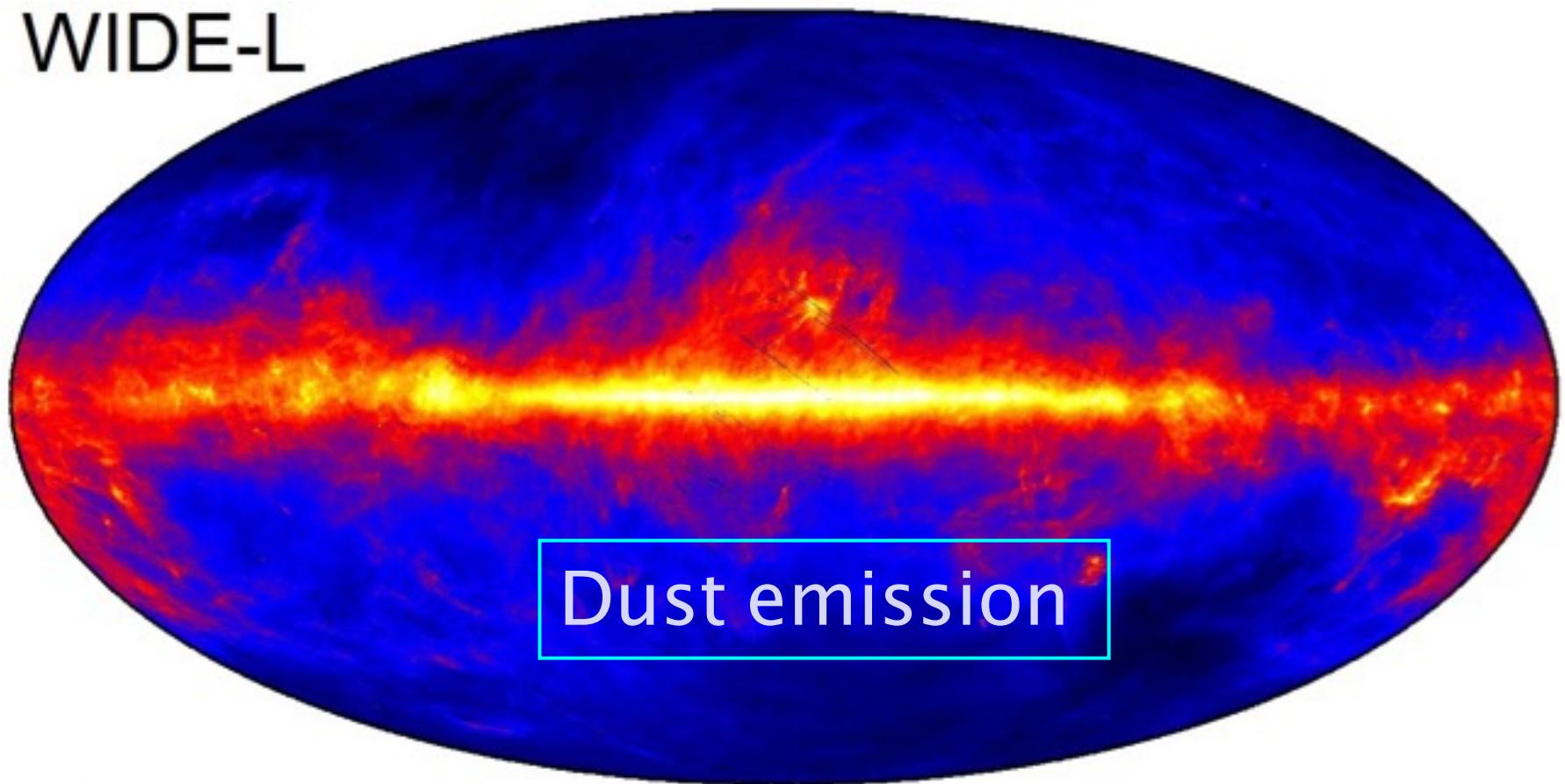
Optically thin radiation reflects:

- (1) column density (ρL)
- (2) optical properties (κ_ν)

Milky Way in the Far Infrared

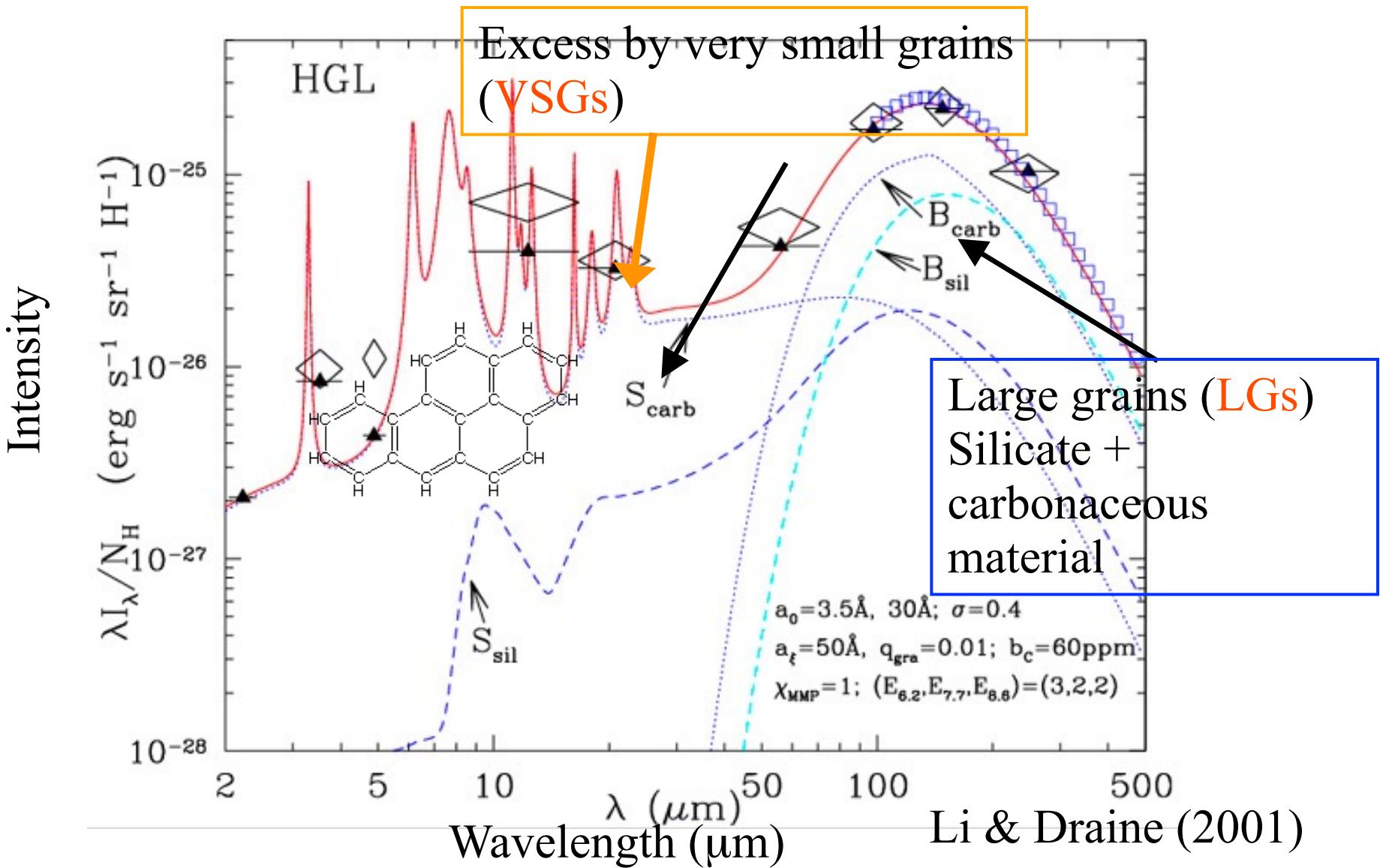
AKARI 140 μm

WIDE-L

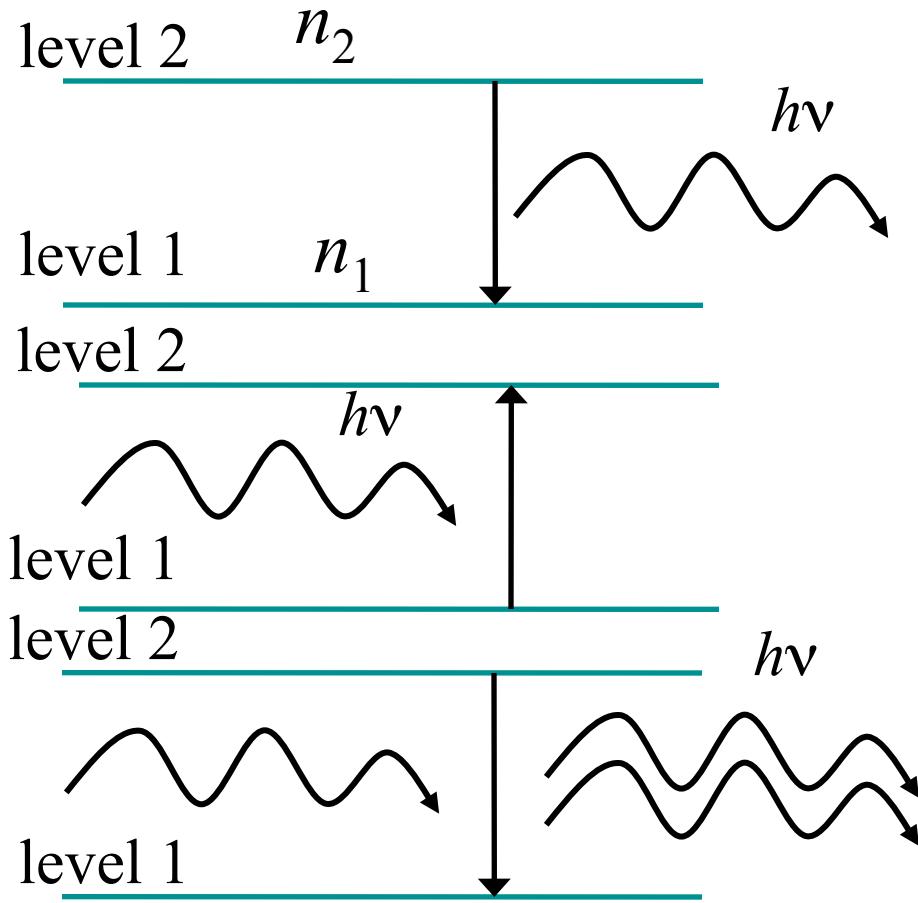


Dust emission

Optical Properties of Dust



4. The Einstein Coefficients



$$J = \int_0^{\infty} J_{\nu} \phi(\nu) d\nu$$

Spontaneous emission

A_{21} (transition probability per unit time)

Absorption

$B_{12}J$ (transition probability per unit time)

Stimulated emission

$B_{21}J$ (transition probability per unit time)

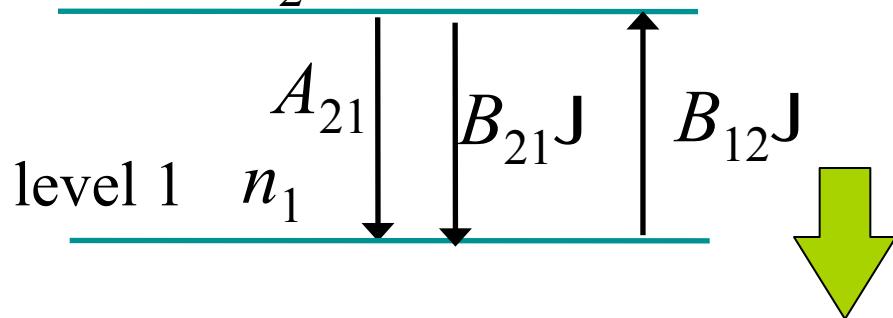
$\phi(\nu)$: line profile function

$$\int_0^{\infty} \phi(\nu) d\nu = 1$$

Einstein Relations

Thermodynamic equilibrium

$$\frac{n_1 B_{12} J}{n_2} = n_2 A_{21} + n_2 B_{21} J \Rightarrow J = \frac{A_{21}/B_{21}}{(n_1/n_2)(B_{12}/B_{21}) - 1}$$



$$\frac{n_1}{n_2} = \frac{g_1}{g_2} \exp(h\nu/kT)$$

$$J = \frac{A_{21}/B_{21}}{(g_1 B_{12}/g_2 B_{21}) \exp(h\nu/kT) - 1}$$

should be the Planck function.

$$\boxed{g_1 B_{12} = g_2 B_{21}}$$
$$A_{21} = (2h\nu^3/c^2)B_{21}$$

Absorption and Emission Coefficients

$$j_\nu = \frac{h\nu}{4\pi} n_2 A_{21} \phi(\nu)$$

$$\alpha_\nu = \frac{h\nu}{4\pi} \phi(\nu) (n_1 B_{12} - n_2 B_{21})$$

$$S_\nu = \frac{n_2 A_{21}}{n_1 B_{12} - n_2 B_{21}}$$

$$(\alpha_\nu = \rho \kappa_\nu)$$

Using the Einstein relations,

$$\alpha_\nu = \frac{h\nu}{4\pi} n_1 B_{12} \left(1 - \frac{g_1 n_2}{g_2 n_1}\right) \phi(\nu)$$

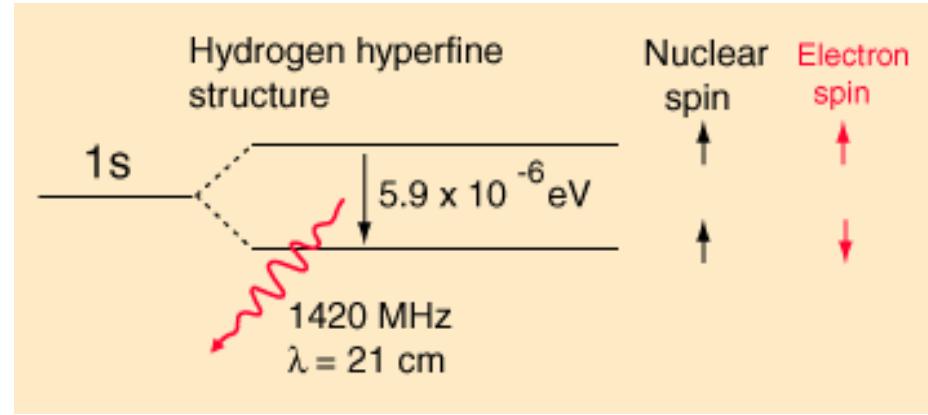
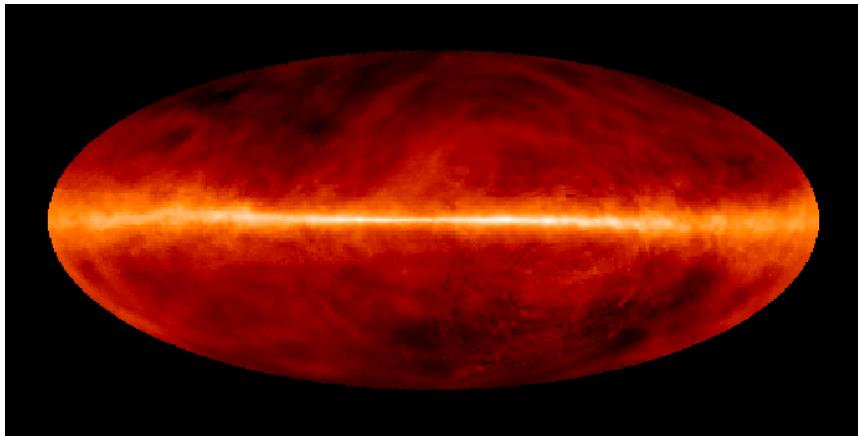
Correction for the
stimulated emission

$$S_\nu = \frac{2h\nu^3}{c^2} \left(\frac{g_2 n_1}{g_1 n_2} - 1 \right)^{-1}$$

For LTE:

$$\alpha_\nu = \frac{h\nu}{4\pi} n_1 B_{12} [1 - \exp(-h\nu/k_B T)] \phi(\nu)$$
$$S_\nu = B_\nu(T)$$

Example for the Correction for the Stimulated Emission: Atomic Hydrogen (21 cm)



$$h\nu/k_B T \ll 1 \rightarrow \alpha_\nu = \frac{h\nu}{4\pi} n_1 B_{12} \frac{h\nu}{k_B T} \phi(\nu) \equiv K_0 n_1 \frac{h\nu}{k_B T} \phi(\nu)$$

$$\tau_\nu = K_0 \frac{N_{\text{HI}}}{4} \frac{h\nu}{k_B T} \phi(\nu) \quad (g_2 = 3, g_1 = 1)$$

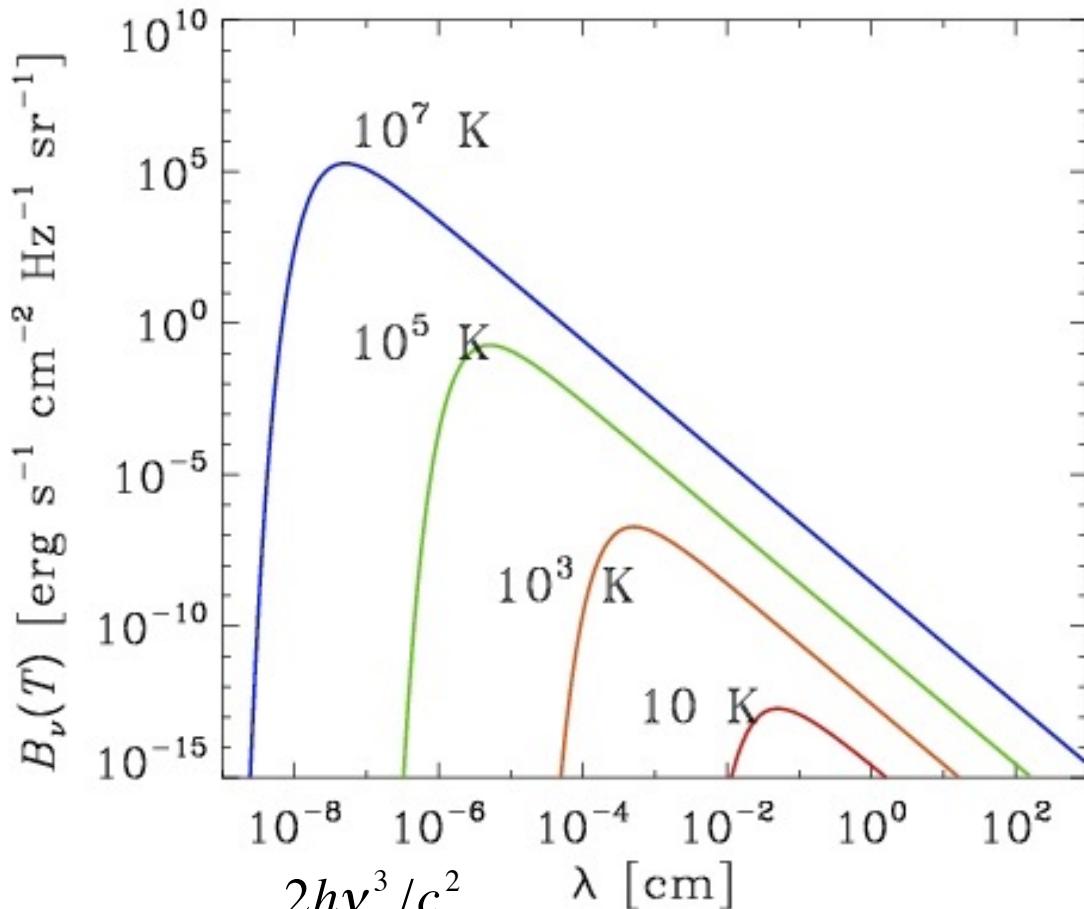
Rayleigh-Jeans Approx.

known constant

$$I_\nu = \tau_\nu B_\nu(T) \simeq \frac{2h\nu^3}{c^2} K_0 \frac{N_{\text{HI}}}{4} \phi(\nu)$$

$$I = \int I_\nu d\nu \simeq \frac{2h\nu^3}{c^2} K_0 \frac{N_{\text{HI}}}{4}$$

Appendix: The Properties of $B_\nu(T)$



$$B_\lambda(T) = \frac{2hc^2/\lambda^5}{\exp(hc/k\lambda T) - 1}$$

- (1) Monotonic increase with T .
- (2) Peak ($\partial B_\lambda / \partial \lambda|_{\lambda=\lambda_{\max}} = 0$): $\lambda_{\max} T = 0.29 \text{ cm deg}$ (Wien displacement law).
- (3) $h\nu \ll kT \Rightarrow B_\nu(T) \sim (2\nu^2/c^2)kT$ (Rayleigh-Jeans law)
- (4) $\int_0^\infty \pi B_\nu(T) d\nu = \sigma T^4$
 $\sigma = 2\pi^5 k^4 / (15c^2 h^3)$
 Stefan-Boltzmann constant

Further Reading

- Rybicki, G. B., & Lightman, A. P. 1979, “Radiative Processes in Astrophysics” (Wiley: New York)