

## Efficient Synthesis of Alkyl End-Capped Oligoheterocycles via the Use of Palladacycle Catalyst

Fen-Tair Luo\* ( 盧鳳泰) and Ashok C. Bajji

*Institute of Chemistry, Academia Sinica, Nankang, Taipei, Taiwan 11529, R.O.C.*

The preparation of mixed thiophene/furan oligomers with alkyl groups at  $\alpha, \alpha'$ -position by the method which we discovered recently in our lab is presented. Thus, the mixed thiophene/furan oligomers can be prepared in good yields from monoiodoarene in the presence of 5 mol % of palladacycle catalyst and 1.2 equiv of *N,N*-diisopropylethylamine in DMF at 100 °C for 8 h.

### INTRODUCTION

Thiophene oligomers have continually attracted much attention to their biological activities<sup>1,2</sup> and as starting materials for the preparation of organic conductors.<sup>3,4</sup> The oligomers of similar structure such as pyrroles and thiophenes are reported and are promising materials for many potential applications.<sup>5,6</sup> However, despite the interesting properties in this field, not many efforts have been undertaken to synthesize the mixed thiophene/furan oligomers with alkyl groups capped at  $\alpha$ -position. The presence of the furan unit should enhance the solubility of oligomers compared with their thiophene analogues. Until recently, very few mixed thiophene/furan oligomers have been reported in the literature.<sup>7-13</sup> The utility of oligomers with well defined structures has become apparent and a number of these oligomers, which are usually  $\beta$ -alkylated to enhance solubility, also have been reported recently.<sup>14-19</sup> The  $\alpha$ -alkyl end-capped oligomers are particularly useful for obtaining the information on the nature of the charge carriers in the doped polythiophene, since blocking of the  $\alpha$ -positions prevents further undesired reactions upon doping. To our knowledge, there are no reports in the literature about the  $\alpha$ -alkyl end-capped mixed oligomers containing thiophene and furan units. In this paper, we wish to report an efficient synthesis of mixed thiophene/furan oligomers with alkyl groups capped at  $\alpha$ -position by the reductive homocoupling method by palladacycle catalyst which we have discovered in our laboratory recently.<sup>20,21</sup>

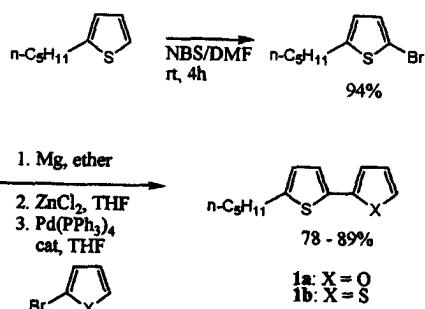
### RESULTS AND DISCUSSION

The synthetic routes used to build up oligoheterocycles

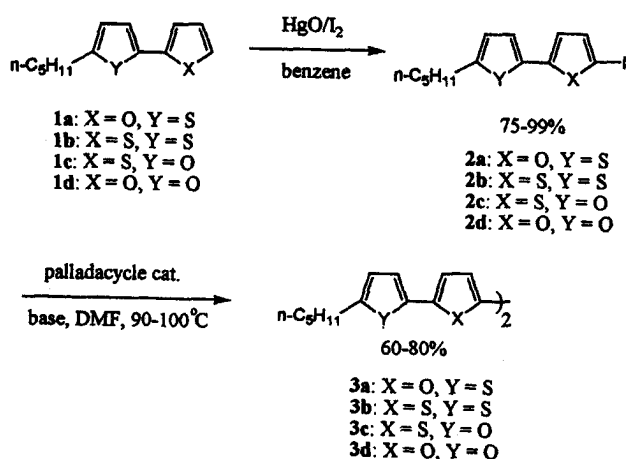
were the homocoupling of the monoiodoheteroaryl by the use of palladacycle catalyst.<sup>20</sup> The preparation of compounds **1a-b** was in satisfactory yields (73 to 83%) from 2-pentylthiophene by bromination (NBS/DMF), transformed into Grignard and zinc reagents, and followed by crossed coupling with 2-bromofuran or 2-bromothiophene by the aid of  $\text{Pd}(\text{PPh}_3)_4$  catalyst as shown in Scheme I. It is noteworthy that the palladium-catalyzed cross coupling reactions of 2-bromo-5-pentylthiophene with either the corresponding Grignard or organozinc compound of 2-bromofuran and 2-bromothiophene failed. Compounds **1c-d** were synthesized in 78 to 85% yields according to our earlier reports.<sup>21</sup> Thus, (3*Z*)-1-(2-furyl)-3-iodonon-3-en-1-one and (3*Z*)-3-iodo-1-(2-thienyl)non-3-en-1-one were prepared from 1-(2-furyl)non-2-yn-1-one and 1-(2-thienyl)non-2-yn-1-one, respectively, by treatment with  $\text{TMSCl}/\text{NaI}/\text{H}_2\text{O}$  in MeCN.<sup>22,23</sup> The following cyclization by the use of palladacycle catalyst and *N,N*-diisopropylethylamine as a base can afford compounds **1c-d**. The products **3a-d** were synthesized from **1a-d** via iodination ( $\text{HgO}/\text{I}_2$ ) in benzene to give **2a-d** (75 to 99% yields) followed by reductive homocoupling reaction in the presence of 5 mol % of palladacycle catalyst and *N,N*-diisopropylethylamine in dry DMF at 90-100 °C (60 to 81% yields) as shown in Scheme II. The use of  $\text{Pd}(\text{PPh}_3)_4$  as the catalyst in the reductive homocoupling reaction gave only low yields (< 25%). The use of the corresponding monobromoarene or monochloroarene in the above reductive homocoupling reaction failed.

The electrochemical and conducting properties of the  $\alpha$ -alkyl end-capped mixed oligomers containing thiophene and furan units are currently under active investigation.

Scheme I



Scheme II



## EXPERIMENTAL SECTION

Precoated silica gel 60F-254 on aluminum plates made by EM chemical company was used for thin-layer chromatography. Purification by column chromatography was carried out with EM silica gel 60 (70-230 mesh ASTM). High-pressure liquid chromatography (HPLC) separation was performed at a flow rate of 0.7 mL/min by the use of two Chemco-Pak 10 × 250 column packed with Chemcosorb 5-ODS-H. GLC analyses were performed by a 3.2 × 3.1 column packed with SE-30 (5% on Chemcosorb W). The purity of each compound was judged to be > 95% by GLC, <sup>1</sup>H-NMR or <sup>13</sup>C-NMR spectral analyses. Reactions of organometallic compounds were undertaken in oven- and/or flame-dried glassware. Tetrakis(triphenylphosphine)palladium<sup>24</sup> and palladacycle<sup>25</sup> were prepared by published methods. Zinc chloride was dried before use at > 50 °C and at < 1 mmHg for 2 h. All other materials were used without further purification. IR spectra were recorded on a Perkin-Elmer 882 infrared spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a

Bruker AC200 or AC300 spectrometer, chemical shifts were reported in ppm down field from Me<sub>4</sub>Si. MS spectra were obtained on HP 5971 or Fisons MD800 GC/MS or VG 70-250S spectrometer.

## 2-Bromo-5-pentylthiophene

In the absence of light, a solution of NBS (9.82 g, 55.19 mmol) in DMF (40 mL) was added dropwise to an ice-cooled solution of 2-pentylthiophene (8.52 g, 55.19 mmol) in DMF (40 mL), and the mixture was stirred for 4 h at room temperature. To the reaction mixture, water (100 mL) was added and the organic compound was extracted with ethyl acetate (50 mL × 2). The combined organic extract was washed with water (50 mL) and brine (50 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated under reduced pressure. The residue was chromatographed over silica gel using hexane as the eluent to give the title compound (12.07 g, 94%) as a pale yellow liquid: <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 300 MHz) δ 0.89 (t, *J* = 7 Hz, 3H), 1.27-1.43 (m, 4H), 1.55-1.70 (m, 2H), 2.73 (t, *J* = 7 Hz, 2H), 6.52 (d, *J* = 3.5 Hz, 1H), 6.83 (d, *J* = 3.5 Hz, 1H) ppm; IR (neat) 1464 (m), 1447 (m), 1045 (w), 956 (w), 790 (m) cm<sup>-1</sup>; MS *m/z* 234, 232 (M<sup>+</sup>), 177, 175, 96, 95; HRMS for C<sub>9</sub>H<sub>13</sub>BrS: calcd 231.9921, found 231.9922.

## 5-Pentyl-2-(2-furyl)thiophene (1a)

Magnesium (0.64 g, 26.46 mmol) turnings and diethyl ether (5 mL) were placed in a flask under nitrogen atmosphere. A crystal of iodine was added to the mixture. Then, 2-bromo-5-pentylthiophene (5.14 g, 22.05 mmol) in diethyl ether (10 mL) was added dropwise at 50 °C. The mixture was refluxed for 2 h and cooled to room temperature at which time ZnCl<sub>2</sub> (3.60 g, 26.46 mmol) in THF (30 mL) was added dropwise and stirred at room temperature for 30 min. To the above reaction mixture 2-bromofuran<sup>26</sup> (3.91 g, 26.46 mmol) and 5 mol % of Pd(PPh<sub>3</sub>)<sub>4</sub> in THF (5 mL) was added dropwise. The reaction mixture was refluxed for 16 h, cooled to room temperature and quenched with saturated NH<sub>4</sub>Cl solution. The organic phase was separated and the aqueous phase was extracted with ethyl acetate (30 mL × 2). The combined organic phase was washed with water (25 mL) and brine (25 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and evaporated under reduced pressure. The residue was chromatographed over silica gel using hexane as the eluent to afford **1a** (3.51 g, 72.31%) as a pale yellow oil: <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 300 MHz) δ 0.89 (t, *J* = 7 Hz, 3H), 1.27-1.43 (m, 4H), 1.66-1.71 (m, 2H), 2.79 (t, *J* = 7 Hz, 2H), 6.41 (s, 2H), 6.69 (d, *J* = 3.5 Hz, 1H), 7.05 (d, *J* = 3.5 Hz, 1H), 7.37 (s, 1H) ppm; <sup>13</sup>C-NMR (CDCl<sub>3</sub>, 300 MHz) δ 13.99, 22.39, 29.99, 31.21, 31.28, 104.12, 111.50, 122.24, 124.55, 131.11, 141.18, 145.13, 149.74 ppm; IR (neat) 1451 (m), 1153 (m), 1012 (m), 974 (m) cm<sup>-1</sup>; MS *m/z* 220 (M<sup>+</sup>), 163,

134, 91; HRMS for  $C_{13}H_{16}OS$ : calcd 220.0922, found 220.0924.

#### 5-Pentyl-2-(2-thienyl)thiophene (1b)

Using the general procedure described above for the synthesis of **1a**, compound **1b** was prepared from 2-bromo-5-pentylthiophene (1.0 g, 4.29 mmol), magnesium (0.13 g, 5.15 mmol), dry ether (10 mL),  $ZnCl_2$  (0.70 g, 5.15 mmol), 5 mol % of  $Pd(PPh_3)_4$  and 2-bromothiophene (0.70 g, 4.28 mmol). After column chromatography over silica gel using hexane as the eluent to provide **1b** (0.89 g, 88.11%):  $^1H$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  0.90 (t,  $J = 7$  Hz, 3H), 1.34-1.38 (m, 2H), 1.63-1.73 (m, 2H), 2.78 (t,  $J = 7$  Hz, 2H), 6.67 (d,  $J = 3.4$  Hz, 1H), 6.97-6.99 (m, 2H), 7.09 (d,  $J = 3.4$  Hz, 1H), 7.15 (d,  $J = 3.4$  Hz, 1H) ppm;  $^{13}C$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  13.98, 22.39, 30.08, 31.25 (2 Cs), 122.93, 123.34, 123.66, 124.66, 127.63, 134.71, 137.94, 145.33 ppm; IR (neat) 1518 (m), 1486 (m), 1206 (m), 837 (m), 795 (s)  $cm^{-1}$ ; MS  $m/z$  236 ( $M^+$ ), 207, 179; HRMS for  $C_{13}H_{16}S_2$ : calcd 236.0693, found 236.0695.

#### 5-Pentyl-2-(2-thienyl)furan (1c)<sup>21</sup>

Pale yellow liquid;  $^1H$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  0.90 (t,  $J = 7$  Hz, 3H), 1.32-1.38 (m, 4H), 1.62-1.72 (m, 2H), 2.64 (t,  $J = 7$  Hz, 2H), 6.01 (d,  $J = 3$  Hz, 1H), 6.38 (d,  $J = 3$  Hz, 1H), 6.98-7.01 (m, 1H), 7.14-7.18 (m, 2H) ppm;  $^{13}C$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  13.99, 22.40, 27.73, 28.04, 31.36, 105.72, 106.72, 121.61, 123.27, 127.48, 134.33, 147.64, 156.13, 185.72 ppm; IR (neat) 3112 (w), 3076 (w), 2949 (s), 2922 (s), 2849 (m), 1560 (m), 1460 (w), 1428 (w), 1374 (w), 1261 (w), 1193 (w), 1012 (m), 844 (m), 772 (s), 681 (s)  $cm^{-1}$ ; MS  $m/z$  220 ( $M^+$ ), 219, 163, 162, 134, 91; HRMS for  $C_{13}H_{16}OS$ : calcd 220.0922, found 220.0922; Anal. Calcd for  $C_{13}H_{16}OS$ : C, 70.87; H, 7.32. Found: C, 70.92; H, 7.50.

#### 2-(2-Furyl)-5-pentylfuran (1d)<sup>21</sup>

Colorless liquid;  $R_f = 0.55$  (Si60 F254, hexane);  $^1H$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  0.89 (t,  $J = 7$  Hz, 3H), 1.31-1.37 (m, 4H), 1.61-1.69 (m, 2H), 2.65 (t,  $J = 7$  Hz, 2H), 6.02 (d,  $J = 2$  Hz, 1H), 6.42-6.47 (m, 3H), 7.38 (s, 1H) ppm;  $^{13}C$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  13.97, 22.39, 27.73, 28.02, 31.35, 104.07, 105.84, 106.45, 111.21, 141.26, 144.82, 146.98, 156.28 ppm; IR (neat) 3115 (w), 2949 (s), 2931 (s), 2859 (m), 1578 (m), 1455 (m), 1202 (w), 1157 (w), 1007 (s), 776 (s), 722 (s)  $cm^{-1}$ ; MS  $m/z$  204 ( $M^+$ ), 147, 95, 91; HRMS for  $C_{13}H_{16}O_2$ : calcd 204.1150, found 204.1150; Anal. Calcd for  $C_{13}H_{16}O_2$ : C, 76.44; H, 7.90. Found: C, 76.66; H, 8.12.

#### 5-Pentyl-2-(5-iodo-2-furyl)thiophene (2a)

To a solution of **1a** (2.00 g, 9.09 mmol) in benzene (15 mL) was added alternately in small portions at room tempera-

ture, mercuric oxide (1.98 g, 9.09 mmol) and iodine (2.31 g, 9.09 mmol). The reaction mixture was stirred at room temperature for 12 h, extracted with ethyl acetate, and washed with  $Na_2SO_3$  aqueous solution. The organic layer was then dried ( $Na_2SO_4$ ), the solvent was removed under reduced pressure, and the residue was chromatographed over silica gel using hexane as the eluent to afford **2a** (2.34 g, 74.47%) as a pale yellow oil:  $^1H$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  0.90 (t,  $J = 7$  Hz, 3H), 1.33-1.40 (m, 4H), 1.57-1.72 (m, 2H), 2.79 (t,  $J = 7$  Hz, 2H), 6.29 (d,  $J = 3.4$  Hz, 1H), 6.55 (d,  $J = 3.4$  Hz, 1H), 6.67 (dt,  $J = 3.6, 1.0$  Hz, 1H), 7.04 (d,  $J = 3.6$  Hz, 1H) ppm;  $^{13}C$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  13.95, 22.37, 29.98, 31.19, 31.25, 85.63, 106.93, 122.12, 122.83, 124.57, 129.91, 145.81, 155.03 ppm; IR (neat) 1638 (m), 1492 (m), 1171 (m), 1105 (m), 1010 (m), 991 (m), 913 (m), 799 (m), 776 (m)  $cm^{-1}$ ; MS  $m/z$  346 ( $M^+$ ), 289, 191, 134; HRMS for  $C_{13}H_{15}IOS$ : calcd 345.9888, found 345.9894.

#### 5-Pentyl-2-(5-iodo-2-thienyl)thiophene (2b)

Following the general procedure described above for the synthesis of **2a**, compound **2b** (0.15 g, 99%) was a colorless solid prepared from **1b** (0.10 g, 0.40 mmol), mercuric oxide (0.09 g, 0.40 mmol) and iodine (0.10 g, 0.40 mmol) in benzene (2 mL):  $^1H$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  0.90 (t,  $J = 7$  Hz, 3H), 1.34-1.37 (m, 4H), 1.60-1.70 (m, 2H), 2.77 (t,  $J = 7$  Hz, 2H), 6.66 (d,  $J = 3.5$  Hz, 1H), 6.76 (d,  $J = 3.5$  Hz, 1H), 6.92 (d,  $J = 3.5$  Hz, 1H), 7.12 (d,  $J = 3.5$  Hz, 1H) ppm;  $^{13}C$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  13.95, 22.37, 30.08, 31.23 (2 Cs), 70.90, 123.87, 124.32, 124.76, 133.50, 137.53, 143.90, 146.00 ppm; MS  $m/z$  362 ( $M^+$ ), 305, 179, 134; HRMS for  $C_{13}H_{15}IS_2$ : calcd. 361.9660, found 361.9660.

#### 5-Pentyl-2-(5-iodo-2-thienyl)furan (2c)

Following the general procedure described above for the synthesis of **2a**, compound **2c** (2.04 g, 80.54%) was prepared as a pale yellow oil from **1c** (1.61 g, 7.32 mmol), mercuric oxide (1.58 g, 7.32 mmol), iodine (1.81 g, 7.32 mmol) and benzene (10 mL):  $^{13}C$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  13.99, 22.38, 27.66, 28.00, 31.33, 70.88, 106.26, 106.84, 122.82, 137.37, 140.20, 146.53, 156.59 ppm; IR (neat) 1560 (m), 1420 (m), 1014 (m), 778 (m)  $cm^{-1}$ ; MS  $m/z$  346 ( $M^+$ ), 289, 163, 134; HRMS for  $C_{13}H_{15}IOS$ : calcd 345.9888, found 345.9897.

#### 5-Pentyl-2-(5-iodo-2-furyl)furan (2d)

Following the general procedure described above for the synthesis of **2a**, compound **2d** (0.17 g, 74.45%) was prepared as a pale yellow oil from **1d** (0.14 g, 0.70 mmol), mercuric oxide (0.15 g, 0.70 mmol), iodine (0.18 g, 0.70 mmol) and benzene (3 mL):  $^1H$ -NMR ( $CDCl_3$ , 300 MHz)  $\delta$  0.83 (t,  $J = 7$  Hz, 3H), 1.20-1.27 (m, 4H), 1.56-1.60 (m, 2H), 2.56 (t,  $J = 7$  Hz,

2H), 5.95 (s, 1H), 6.30 (s, 1H), 6.38 (s, 1H), 6.48 (s, 1H) ppm;  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$  14.00, 22.38, 27.67, 27.98, 31.33, 85.96, 106.57, 106.67, 106.76, 121.86, 143.65, 152.13, 156.63 ppm; MS  $m/z$  330 ( $\text{M}^+$ ), 273, 175, 118; HRMS for  $\text{C}_{13}\text{H}_{15}\text{IO}_2$ : calcd 330.0117, found 330.0118.

#### 5,5'-Bis(5-pentyl-2-thienyl)-2,2'-bifuran (3a)

To a solution of **2a** (0.14 g, 0.41 mmol) and 5 mol % of palladacycle in dry DMF (2 mL) was added *N,N*-diisopropylethylamine (0.06 g, 0.49 mmol). The reaction mixture was stirred under heating at 90 to 100 °C for 8 h. The mixture was cooled to room temperature, water was added and extracted with ethyl acetate (15 mL  $\times$  2). The combined organic extract was washed with water (20 mL), dried ( $\text{Na}_2\text{SO}_4$ ), filtered and concentrated under reduced pressure. The dark brown residue was chromatographed over silica gel using hexane as the eluent to provide **3a** (0.06 g, 62.76%) as a pale yellow solid: mp: 75–76 °C;  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$  0.90 (t,  $J$  = 7 Hz, 6H), 1.30–1.40 (m, 8H), 1.67–1.72 (m, 4H), 2.81 (t,  $J$  = 7 Hz, 4H), 6.48 (d,  $J$  = 3.5 Hz, 2H), 6.62 (d,  $J$  = 3.5 Hz, 2H), 6.71 (d,  $J$  = 3.5 Hz, 2H), 7.11 (d,  $J$  = 3.5 Hz, 2H) ppm;  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$  14.00, 22.40, 30.06, 31.22, 31.31, 106.19, 107.16, 122.48, 124.71, 130.73, 145.02, 145.43, 149.01 ppm; MS  $m/z$  438 ( $\text{M}^+$ ), 381, 324, 219, 162; HRMS for  $\text{C}_{26}\text{H}_{30}\text{O}_2\text{S}_2$ : calcd 438.1687, found 438.1691.

#### 5,5'-Bis(5-pentyl-2-thienyl)-2,2'-bithiophene (3b)

The title compound **3b** (0.07 g, 79.55%) was prepared as a pale yellow solid from **2b** (0.14 g, 0.39 mmol), *N,N*-diisopropylethylamine (0.06 g, 0.46 mmol) and 5 mol % of palladacycle in dry DMF (2 mL): mp: 177–179 °C;  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$  0.91 (t,  $J$  = 7 Hz, 6H), 1.30–1.40 (m, 8H), 1.60–1.70 (m, 4H), 2.79 (t,  $J$  = 7 Hz, 4H), 6.68 (d,  $J$  = 3.5 Hz, 2H), 6.97–7.03 (m, 6H) ppm;  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$  13.99, 22.40, 30.15, 31.25, 123.37, 123.55, 124.00, 124.83, 134.45, 135.35, 136.73, 145.66 ppm; MS  $m/z$  470 ( $\text{M}^+$ ), 413, 356, 323, 202, 122, 93; HRMS for  $\text{C}_{26}\text{H}_{30}\text{S}_4$ : calcd 470.1230, found 470.1217.

#### 5,5'-Bis(5-pentyl-2-furyl)-2,2'-bithiophene (3c)

The title compound (0.03 g, 68.60%) was prepared as a yellow solid from **2c** (0.03 g, 0.09 mmol), *N,N*-diisopropylethylamine and 5 mol % of palladacycle in dry DMF (1 mL): mp: 129–130 °C;  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$  0.91 (t,  $J$  = 7 Hz, 6H), 1.30–1.40 (m, 8H), 1.60–1.70 (m, 4H), 2.65 (t,  $J$  = 7 Hz, 4H), 6.03 (d,  $J$  = 3.5 Hz, 2H), 6.39 (d,  $J$  = 3.5 Hz, 2H), 7.04–7.08 (m, 4H) ppm;  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$  14.01, 22.41, 27.72, 28.08, 31.37, 106.08, 106.98, 122.21, 123.85, 132.80, 135.14, 147.21, 156.45 ppm; MS  $m/z$  438 ( $\text{M}^+$ ), 381, 324, 295, 219, 162, 73; HRMS for  $\text{C}_{26}\text{H}_{30}\text{O}_2\text{S}_2$ : calcd

438.1687, found 438.1692.

#### 5,5'-Bis(5-pentyl-2-furyl)-2,2'-bifuran (3d)

The title compound (0.09 g, 59.35%) was prepared as a yellow solid from **2d** (0.25 g, 0.76 mmol), *N,N*-diisopropylethylamine (0.12 g, 0.92 mmol) and 5 mol % of palladacycle in dry DMF (2 mL): mp: 71–73 °C;  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$  0.91 (t,  $J$  = 7 Hz, 6H), 1.33–1.39 (m, 8H), 1.63–1.70 (m, 4H), 2.60–2.72 (m, 4H), 6.05 (d,  $J$  = 3.5 Hz, 2H), 6.45–6.65 (m, 6H) ppm;  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$  14.00, 22.40, 27.71, 28.07, 31.37, 106.05, 106.28, 106.67, 106.98, 144.51, 145.10, 146.20, 156.64 ppm; MS  $m/z$  406 ( $\text{M}^+$ ), 349, 292, 203, 146; HRMS for  $\text{C}_{26}\text{H}_{30}\text{O}_4$ : calcd 406.2144, found 406.2141.

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#### Key Words

Palladacycle catalyst; Mixed furan/thiophene oligomers; Homocoupling reaction; Palladium catalyst.

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