



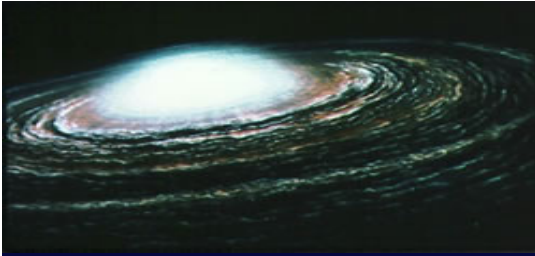
**The Meteoritic Record:
Tracing Early Solar System Evolution**

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Meteorite analysis



- Asteroid fragments - “meteorites” are the only extraterrestrial material that can be investigated in the laboratory in substantial quantities
- The analyses of meteorites provide evidence that both volatile and refractory material has been substantially altered owing to thermal and aqueous processes within the solar system
- Chemical characterization and stable isotope measurements on individual molecules are revealing reaction mechanisms that preceded and coincided with the birth of the solar system



Solar nebula carbon chemistry

- **Infalling interstellar material experiences chemical alteration to varying degrees, depending upon the epoch and position of entry into the nebula**

Ionization source: **stellar X-rays, stellar UV, interstellar UV, cosmic rays**

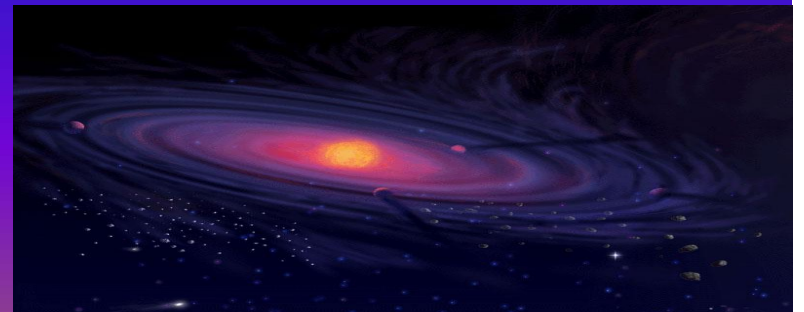
Heating: **viscous dissipation, stellar radiation**

Transport: **radial mixing, diffusion**

- **Inner solar nebula: material is heavily processed, strong radiation, high temperature, dissociation of molecules**
- **Outer solar nebula: evaporation-recondensation of ices, similar chemistry as in dense clouds**

Solar System Inventory

- **Some relative pristine material with a strong interstellar heritage**
- **Highly processed interstellar material that was exposed to high temperatures and radiation**
- **Material newly formed in the solar nebula or on solid solar system objects**



Minerals in meteorites

Chondrites

- Mineral components
 - chondrules
 - clays
 - carbonates
- Multiple proposed origins
 - circumstellar
 - carbonates
 - solar nebula
 - chondrules
 - clays
 - parent body
 - carbonates
 - clays



Axtell CV3

Lithium in meteorites

Lithium

- Two isotopes
 - ${}^6\text{Li}$, ${}^7\text{Li}$
- Action of water
 - ${}^7\text{Li}$ passes preferentially into solution
- Reveals origins of minerals
 - chondrules (anhydrous)
 - ${}^7\text{Li}$ -poor
 - solar nebula
 - clays
 - ${}^7\text{Li}$ -intermediate
 - some water on parent body
 - carbonates
 - ${}^7\text{Li}$ -rich
 - lots of water on parent body



Murchison CM2

Sample	${}^7\text{Li}/{}^6\text{Li}$	$\delta^7\text{Li}^a$
Whole rock.....	12.0654	+3.78 \pm 0.21
Carbonate-rich acetic acid-soluble phase ^b	12.1708	+12.55 \pm 0.17
	12.1766	+13.03 \pm 0.25
Phyllosilicate-rich matrix ^c	12.0940	+6.16 \pm 0.78
	12.0901	+5.83 \pm 0.61
Chondrule.....	11.9969	-1.92 \pm 0.31

Sephton et al. 2004 ApJ 612, 588–591

Carbon in primitive meteorites

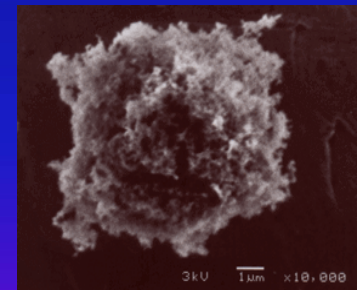
	Abundance (wt%)	$\delta^{13}\text{C}$ (‰)
Whole rock	-	0 to -25
Organic matter	2.0	-13 to -21
Carbonate	0.2	+20 to +80
Diamond	0.04	-38
Graphite	0.005	-50 to +340
Silicon carbide	0.009	+1200

Carbonaceous chondrites – most pristine!



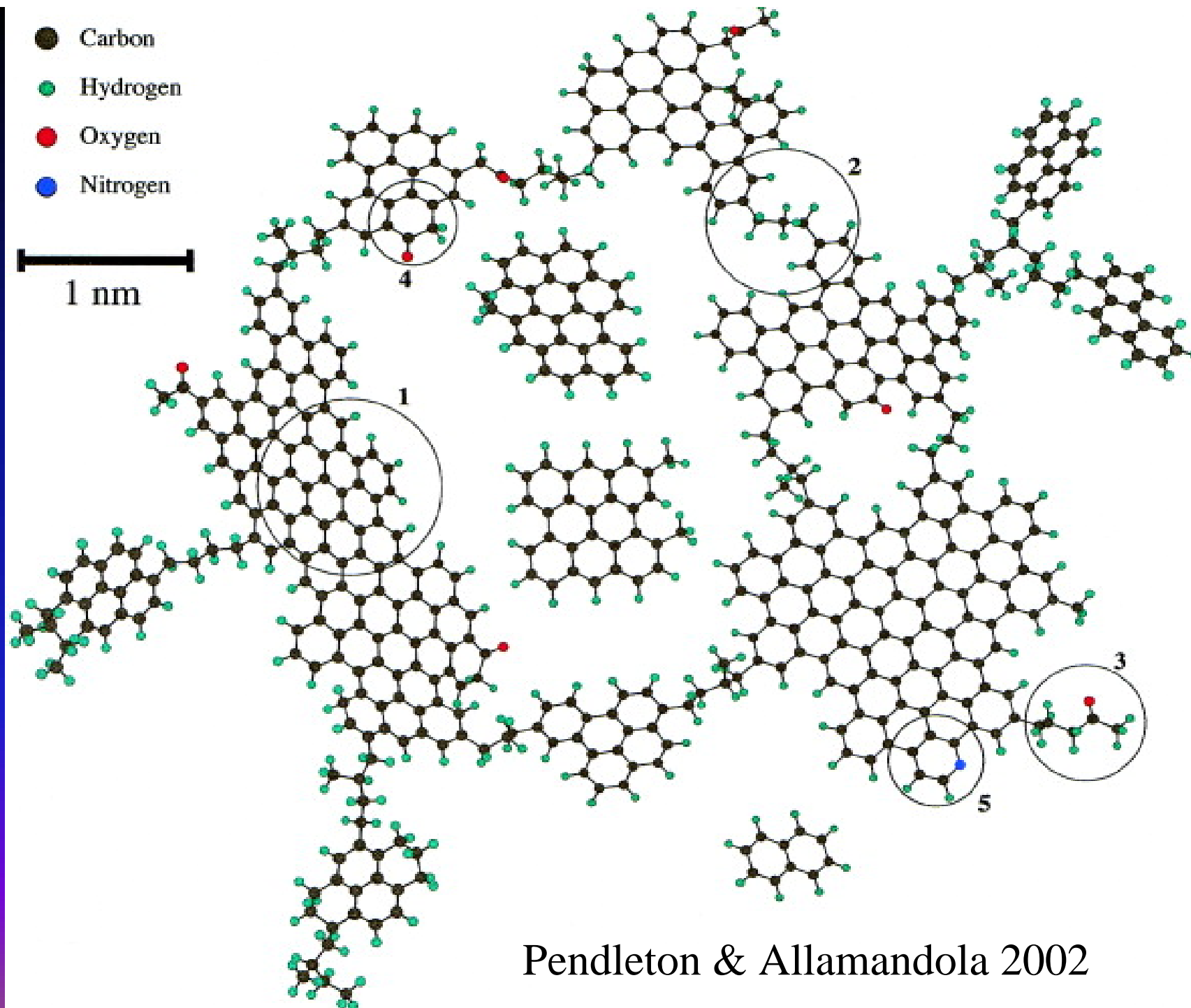
Murchison

Insoluble Carbon-fraction:
60-80 % aromatic carbon
highly substituted small
aromatic moieties branched
by aliphatic chains



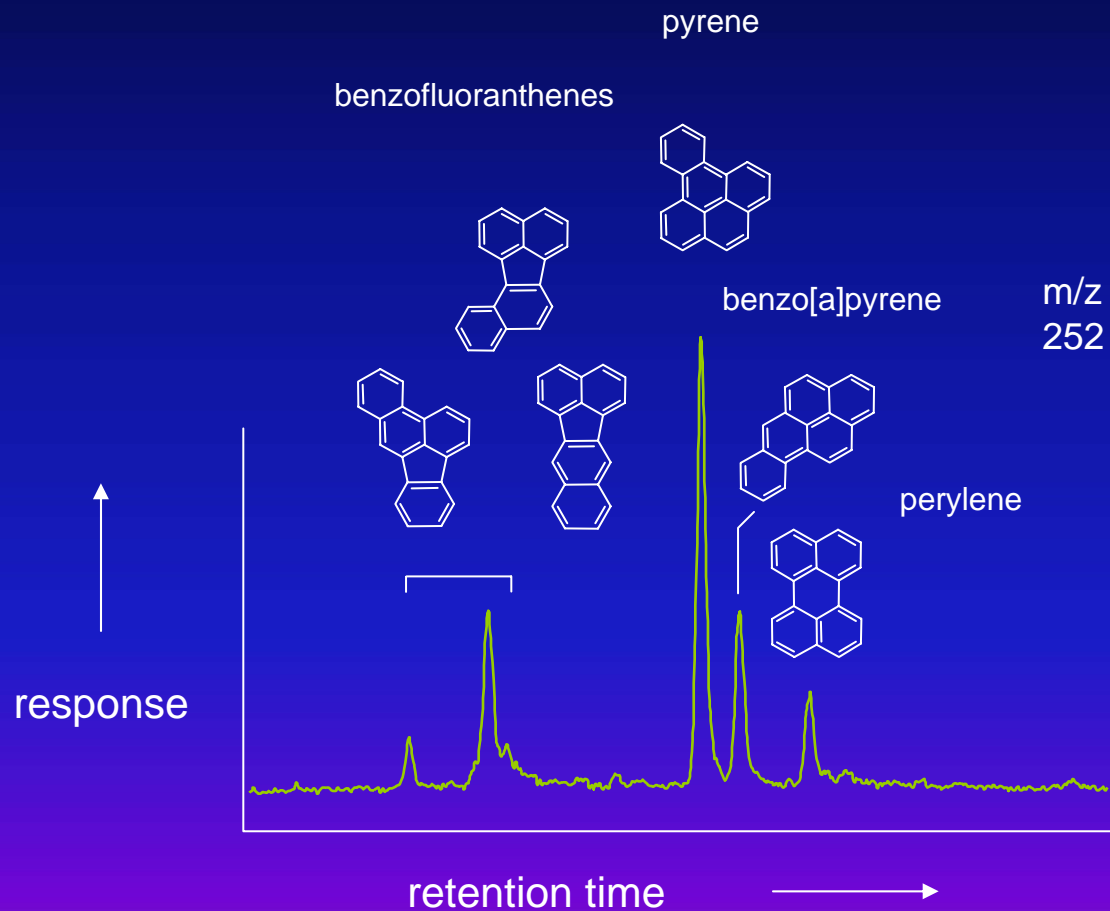
Volatile fraction:





Pendleton & Allamandola 2002

Origin of hydropyrolysate PAH



Ion chromatogram

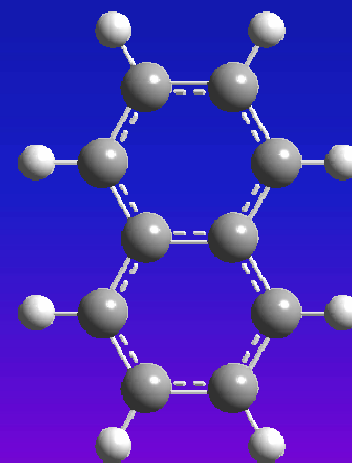
Parental PAH

- poorly cross-linked units
 - short aliphatic chains
 - heteroatoms
- thermodynamics
 - unfavourable forms present
- pre-incorporation?
 - low temperature
 - isotopic heterogeneity
- interstellar?

Sephton et al. 2004

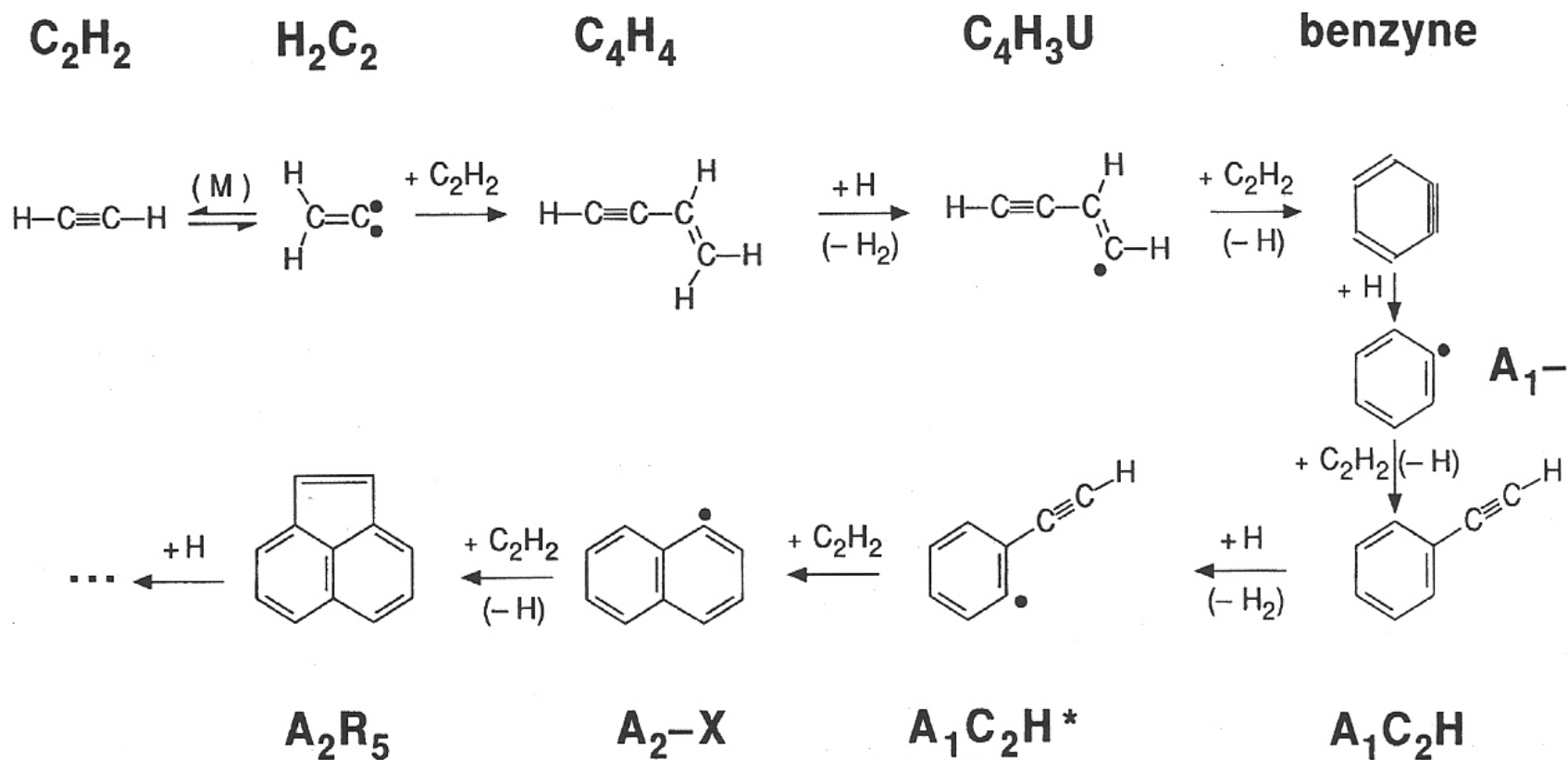
A hot or cold origin for aromatics ?

- **D-enrichment, hotspots ^{13}C , ^{15}N**
- **Increase in $\delta^{13}\text{C}$ and then decrease with increasing molecular size**
- **Large fractionation in C isotopes (in Murchison -13.1 and -5.9 ‰ for pyrene and fluoranthene respectively)**



The dominant reaction pathway of polycyclic aromatic hydrocarbons in circumstellar envelopes

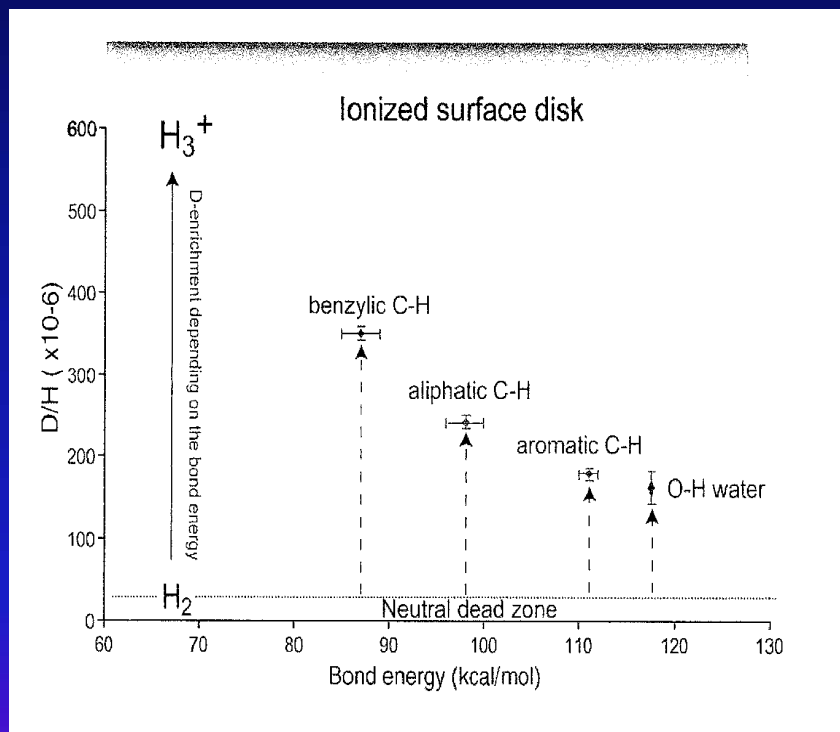
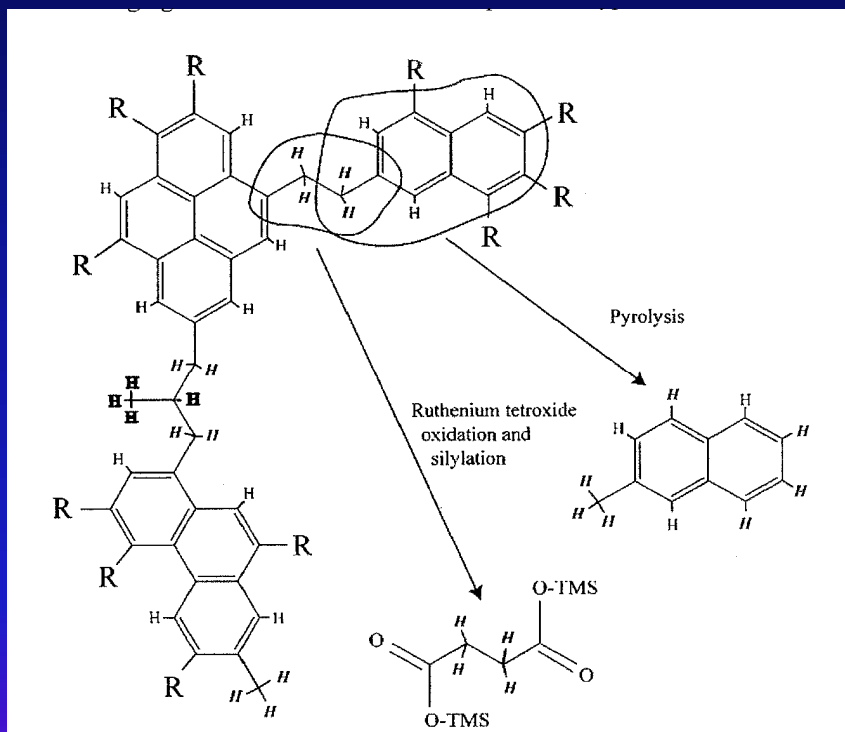
H abstraction, C₂H₂ addition, PAHs form at high T (900-1100 K)



Frenklach & Feigelson 1989

A solar system origin for D-enrichment in organic matter

(Remusat et al. 2006)



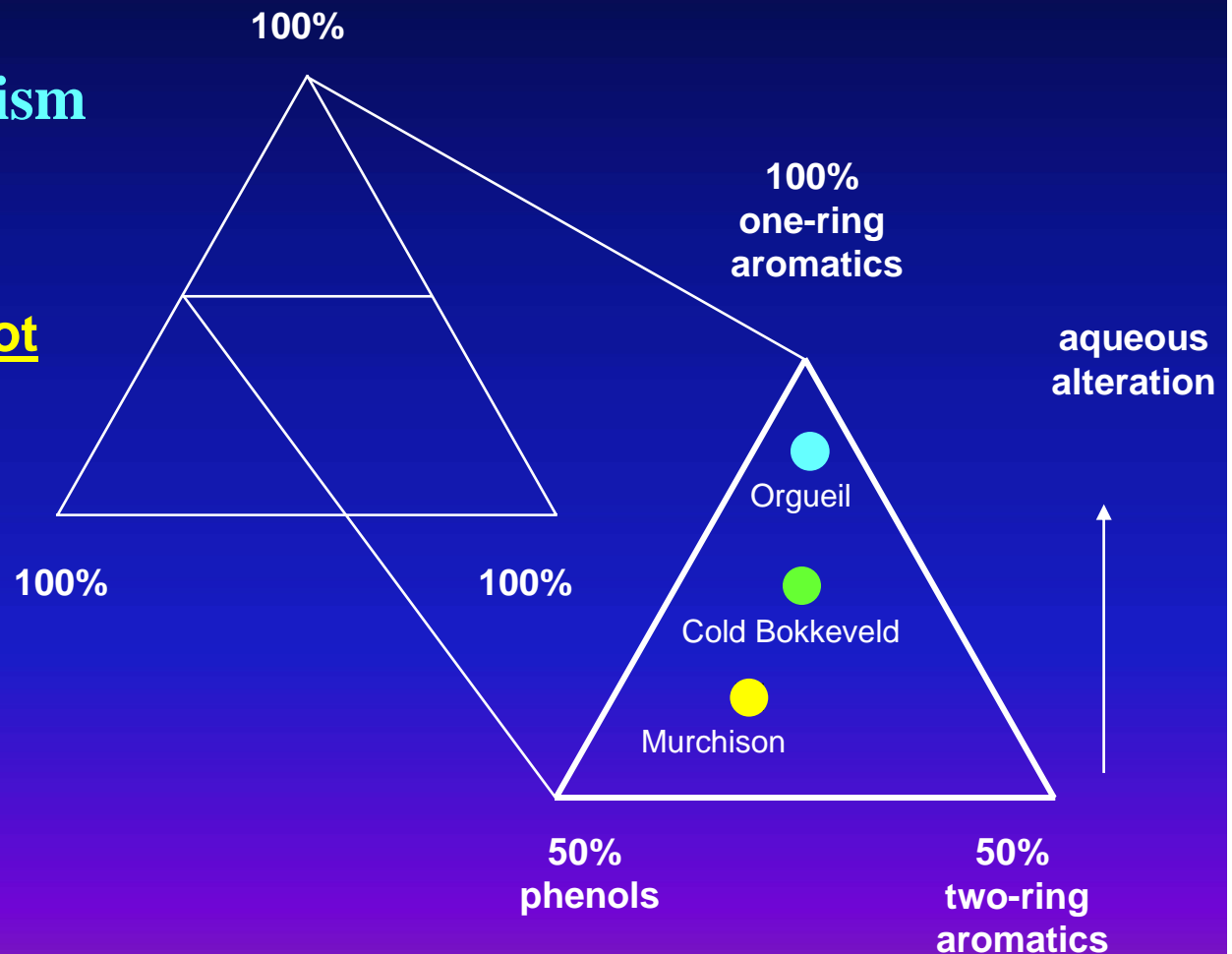
D/H ratio of benzylic, aliphatic and aromatic hydrogen and water as a function of the C-H bond energy. The water D/H indicates that this relation may illustrate a common process in the early solar system.

Quantitative relationships

Aqueous alteration
Oxidation
Thermal Metamorphism

Pyrolysates - ternary plot

- quantitative differences
 - organic moieties
- one ring/two ring
 - degree of condensation
- hydrocarbons/phenols
 - oxygen content
- correlation
 - aqueous alteration

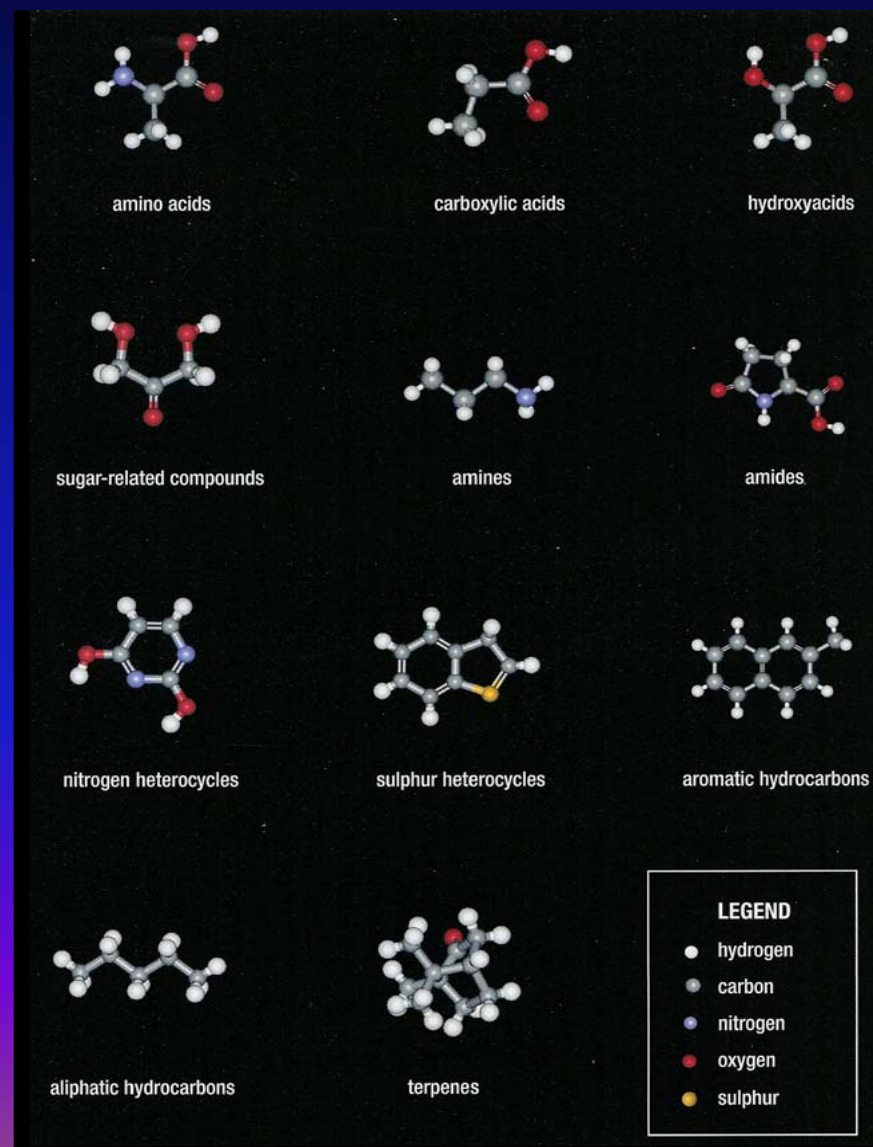


Organic compounds in the Murchison meteorite

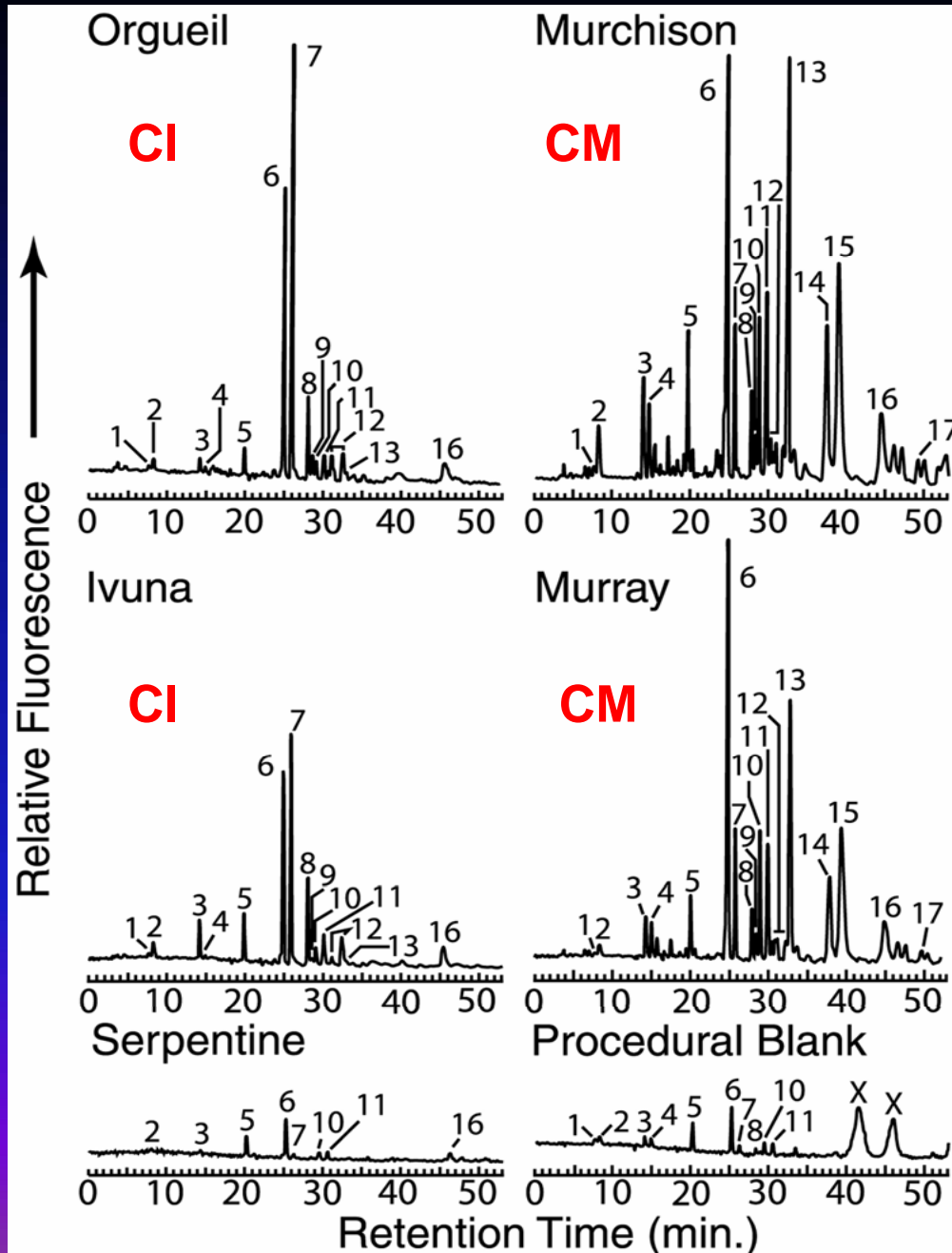
Compound Class Concentration(ppm)

Sephton 2002

CO₂	106
CO	0.06
CH₄	0.14
NH₃	19
Aliphatic hydrocarbons	12-35
Aromatic hydrocarbons	15-28
Amino Acids	60
Monocarboxylic acids	332
Dicarboxylic acids	26
α-hydroxycarboxylic acids	14
Polyols (sugar-related)	~24
Basic N-heterocycles	0.05-0.5
Purines	1.2
Pyrimidines	0.06
Amines	8
Urea	25
Benzothiophenes	0.3
Alcohols	11
Aldehydes	11
Ketones	16

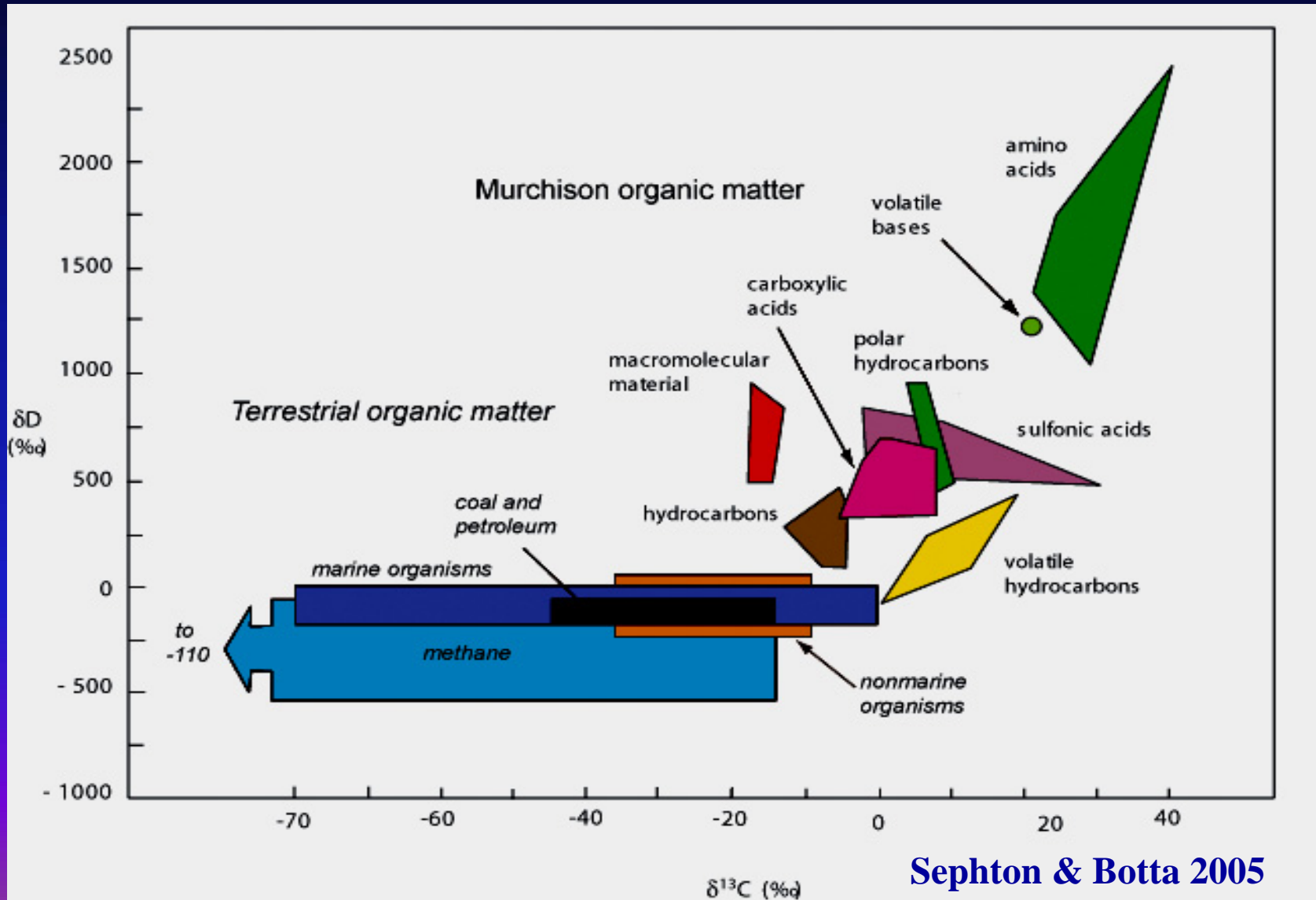


Amino Acids in Meteorite Extracts



- 1 D-Aspartic Acid
- 2 L-Aspartic Acid
- 3 L-Glutamic Acid
- 4 D-Glutamic Acid
- 5 D,L-Serine
- 6 Glycine
- 7 β -Alanine
- 8 γ -Amino-*n*-butyric Acid (γ -ABA)
- 9 D,L- β -Aminoisobutyric Acid (β -AIB)
- 10 D-Alanine
- 11 L-Alanine
- 12 D,L- β -Amino-*n*-butyric Acid (β -ABA)
- 13 α -Aminoisobutyric Acid (AIB)
- 14 D,L- α -Amino-*n*-butyric Acid (α -ABA)
- 15 D,L-Isovaline
- 16 L-Valine
- 17 D-Valine
- X: unknown

Distinction between stable carbon and hydrogen isotope ratios in Murchison and life



Molecule	Hale-Bopp	ISM
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H ₂ O	100	100
CO	12-23	1-60
CO ₂	6	15-40
CH ₄	1.5	
C ₂ H ₂	0.1-0.3	
C ₂ H ₆	0.6	
CH ₃ OH	2.4	1-25
H ₂ CO	1.1	
HCOOH	0.09	
HCOOCH ₃	0.08	
CH ₃ CHO	0.02	
NH ₂ CHO	0.015	
HCN	0.25	
HNCO	0.10	
HNC	0.04	
CH ₃ CN	0.02	
HC ₃ N	0.02	
OCS	0.4	
CS ₂	0.2	
H ₂ CS	0.05	



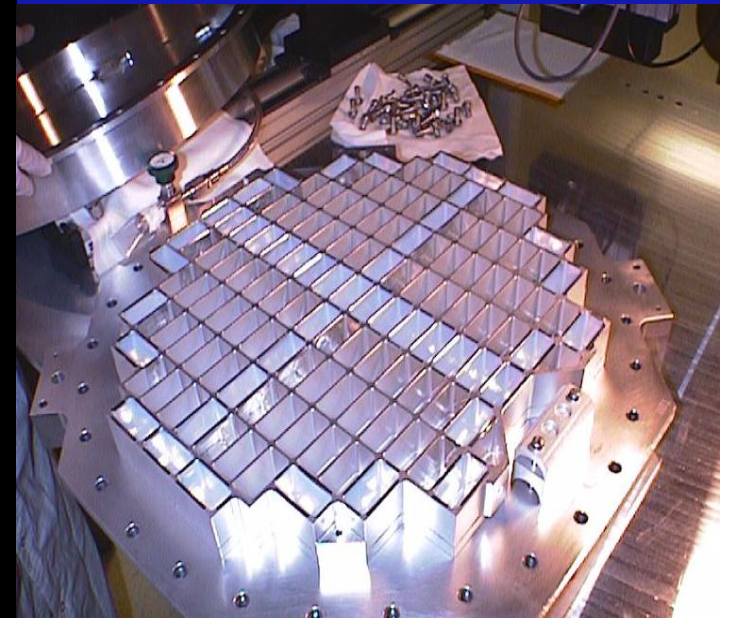
Production rates relative to water
of organic molecules in the coma of
comet

C/1995 O1 Hale-Bopp

(Bockelee-Morvan et al. 2004,
Ehrenfreund et al. 2004)

FLYBY Comet Wild 2

Capture of particles



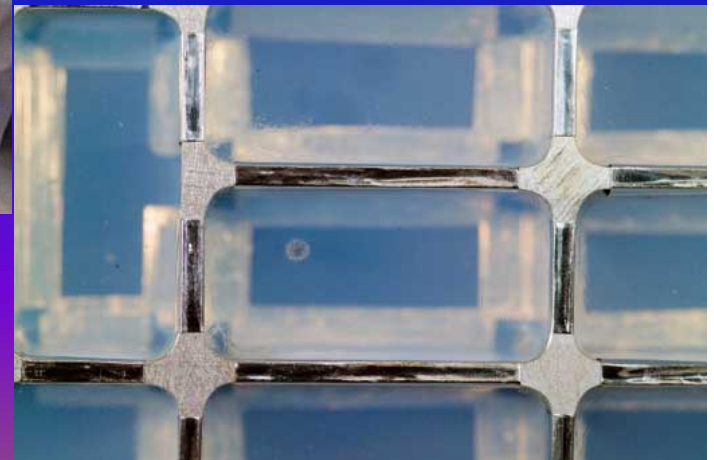
Stardust touch down 15 January 2006 in Utah

Particles from a comet and interstellar dust have been returned to Earth

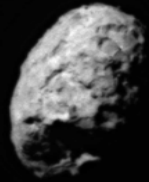


Silicon-based solid 1,000 times less dense than glass

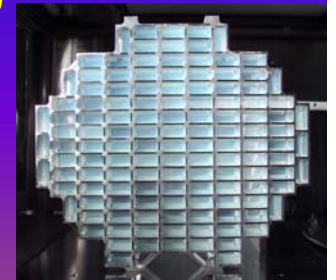
Thousands of impacts on the aerogel



Stardust clues...

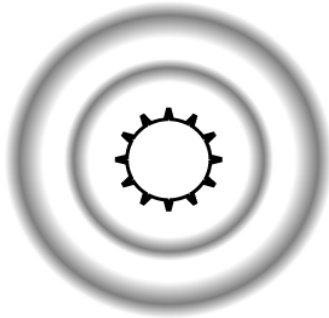


- **Comet Wild 2 dust was predominantly material from our solar nebula – limited amount of preserved presolar material (isotopic O, McKeegan et al. 2006) – 1 CAI**
- **Significant radial mixing in the early solar system**
- **Widely varying olivine and pyroxene compositions - wide range of formation conditions/locations, no aqueous alteration (Zolensky et al. 2006)**

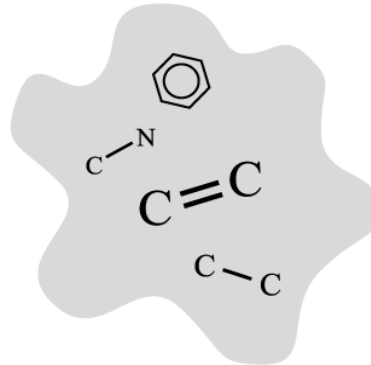


Chemistry in space

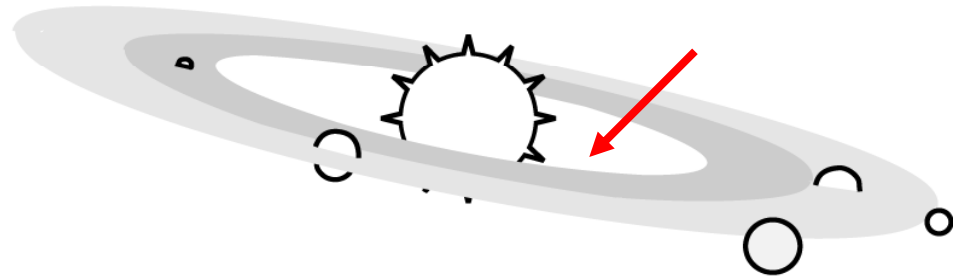
Complex carbon chemistry
in circumstellar envelopes



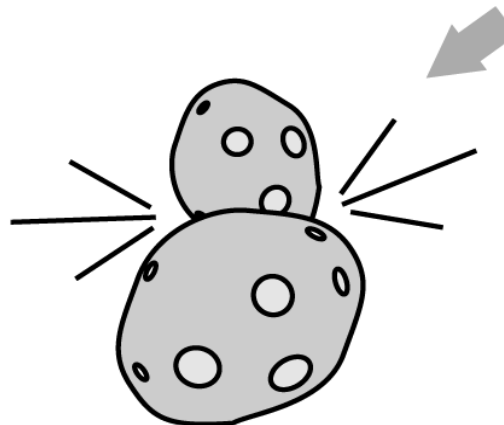
Organic chemistry in
interstellar clouds



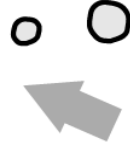
Star systems condense,
chemically seeded



Asteroids and comets
slowly distribute matter



Meteoritic infall
of organic matter



Summary

➤ **Meteorites deliver a protracted record of chemical evolution and links between interstellar and solar system material**

→ **constraints on the physical and dynamical properties of early solar system formation**

➤ **Extensive mixing in the solar nebula**

➤ **Active solar nebula chemistry**

➤ **Similar processes in extrasolar systems**

